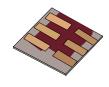
Solar cells and solar energy





Dr. Roderick MacKenzie

Overview



- Why Solar energy?
- Sunlight
- Absorbing sunlight in materials.
- Fundamentals of diodes
- From diodes to solar cells
- Diodes current-voltage curves in the light
- Different types of solar cells
 - Silicon solar cells
 - Organic solar cells
 - Multi-junction solar cells
 - Perovskite solar cells
 - Cadmium telluride solar cells
 - Concentrator solar cells
- Summary

Aims of lecture



•What this lecture is:

- •This lecture aims to give you enough information to understand the operation of solar cells.
- •It is aimed at helping you understand the devices enough to make component choices when designing systems.

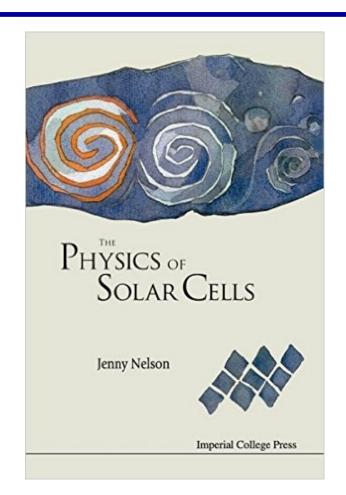
•What this lecture is not:

•A solid state physics lecture.

•This lecture was originally given to an MSc class at the University of Nottingham in 2016.

Recommended book





- •This lecture is only 1.5 hours long, so I can't cover everything about solar cells and solar energy.
- The Physics of Solar Cells (Properties of Semiconductor Materials)
- •If you want a deeper understanding get this book.

Jenny Nelson

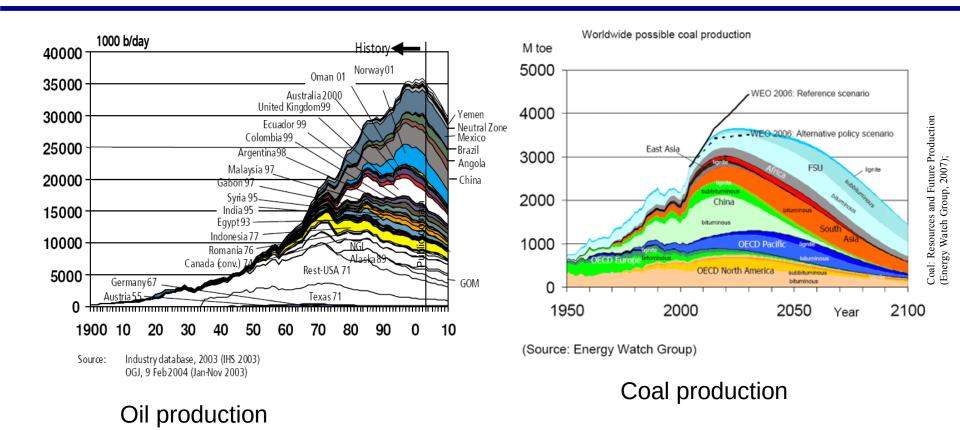
ISBN:1860943497

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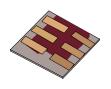
Reason 1: We are running out of oil and coal

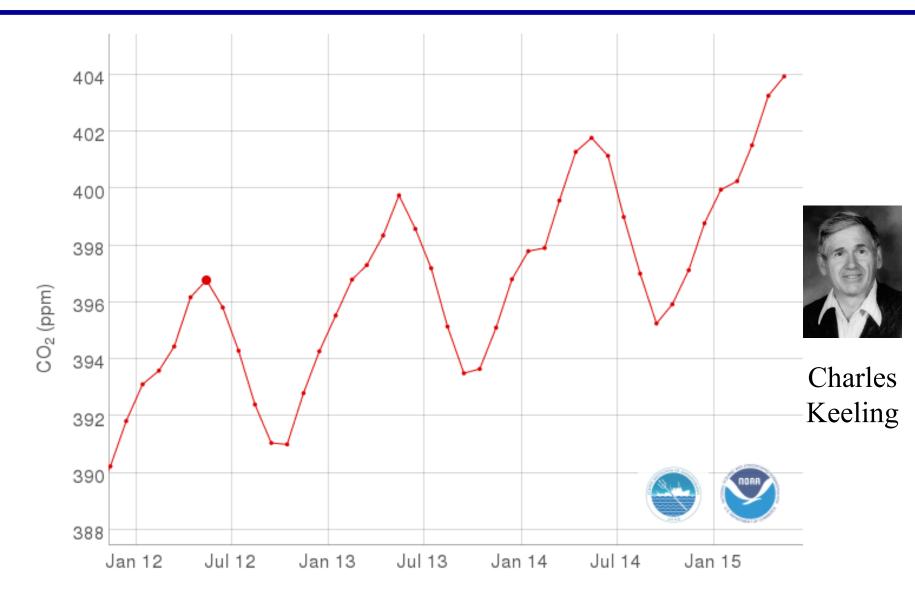


•This will damage the economy and our standard of living, our health and well being.

Nature, January 2012, Vol 481, pp. 433-435

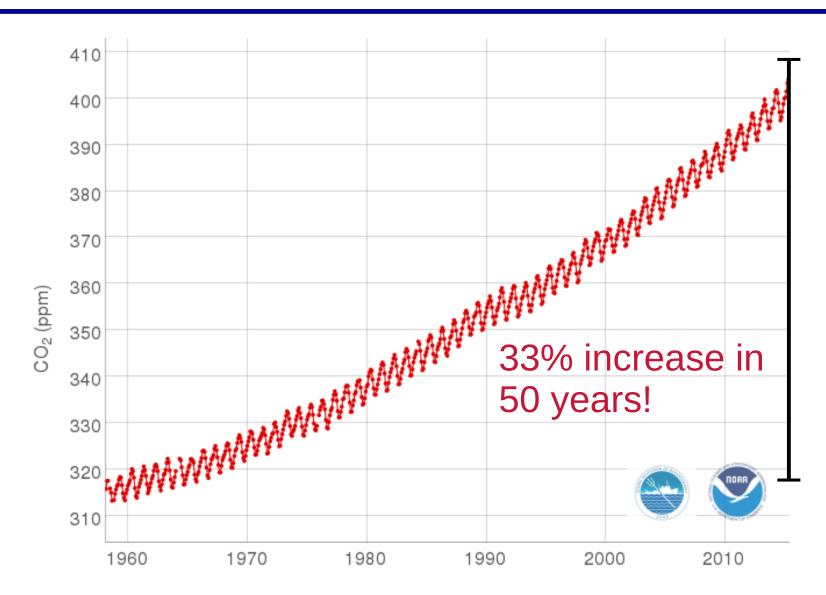
Reason 2: Global warming



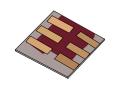


Jan 1980-June 2015

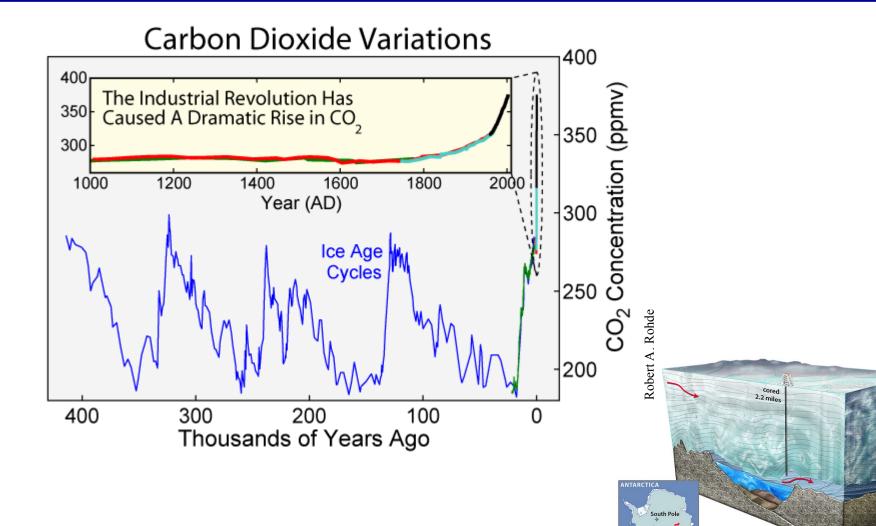




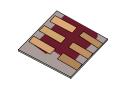
Carbon dioxide 400 AD - 2009 AD

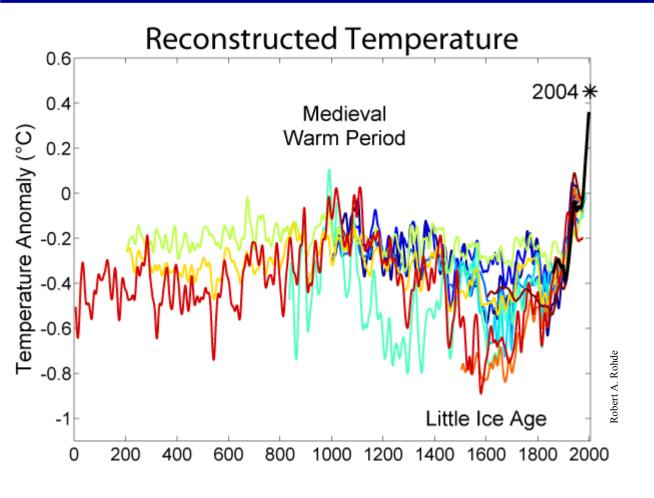


Nicolle Rager-Fuller / NSF



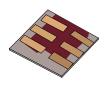
Global temperature 1000 AD - 2000 AD

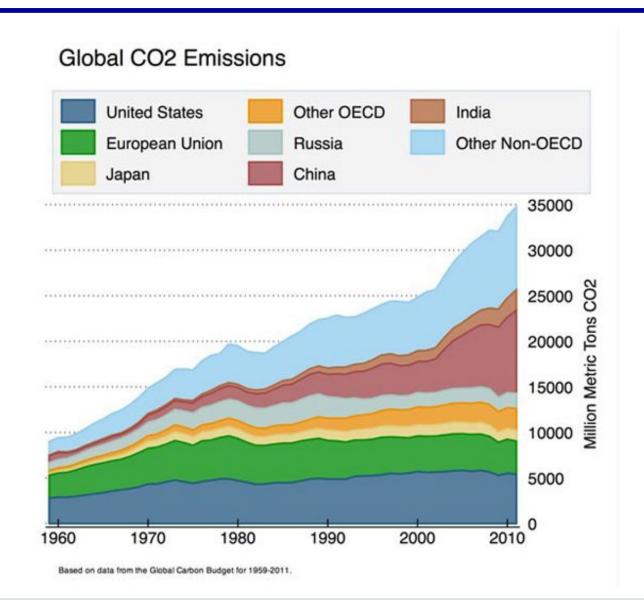






CO₂ Emissions by country

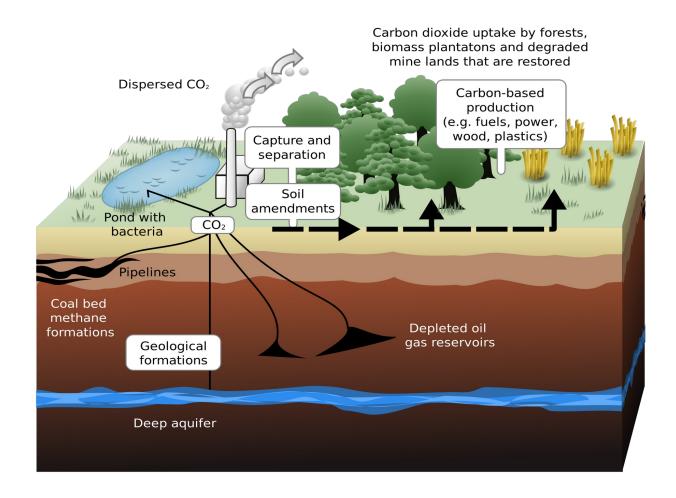




Carbon capture a silly idea.



- Transport CO2
- Compress CO2



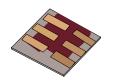
Hydrogen cars?

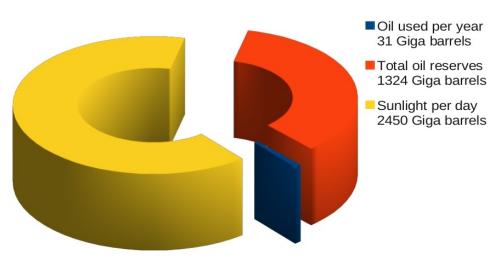




Anyone see a problem?

Why choose solar energy?



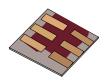


- •Solar max power flux: ~1500 W / m²
- •Average density over the year:
 - •Sahara: ~ 400 W / m²
 - •UK: ~ 100 W / m²

- •Typical solar cell efficiency 15%
- •In UK need 40 m² per person to supply average electricity demand (700 W)



Where do we find PV systems?





Andrewglaser



Sawu12

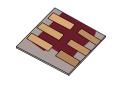
Building integrated

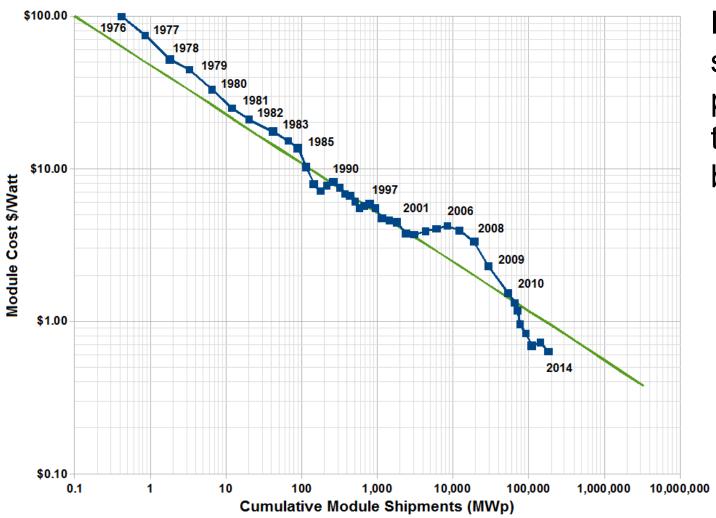
Solar power stations



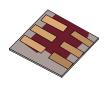
Transport integrated??

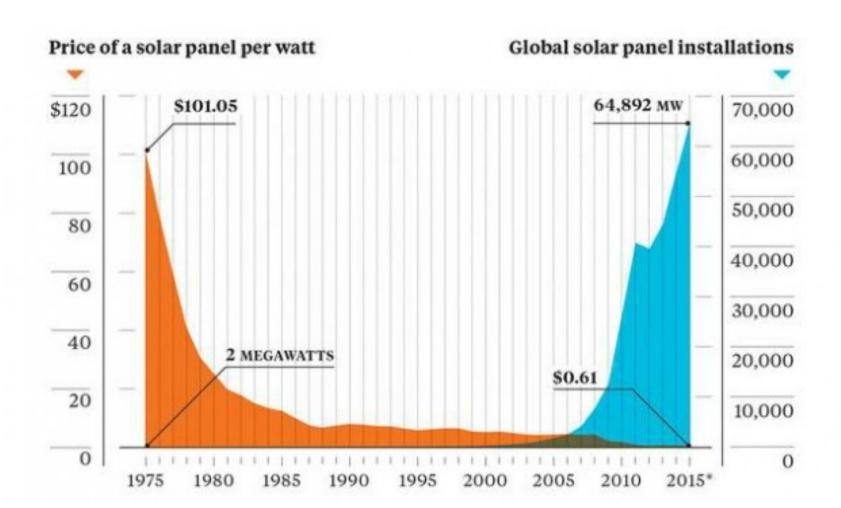
Swanson's law



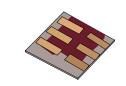


Every time solar cell production the price falls by 20%.

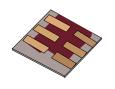


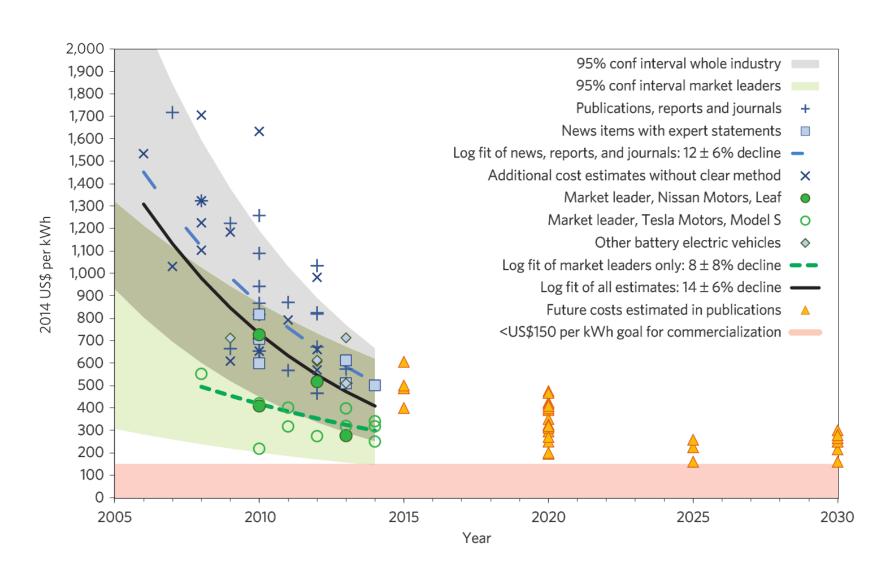


Video

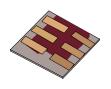


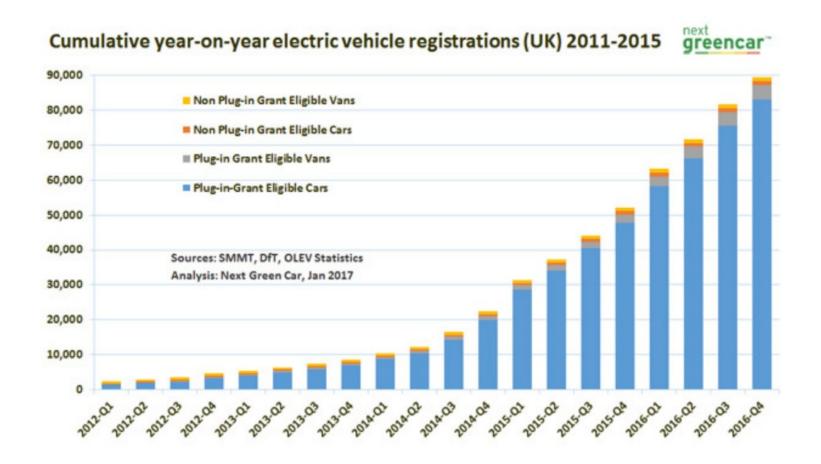
Cost of Li batteries



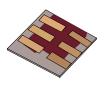


EV car registrations





Cell production





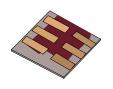


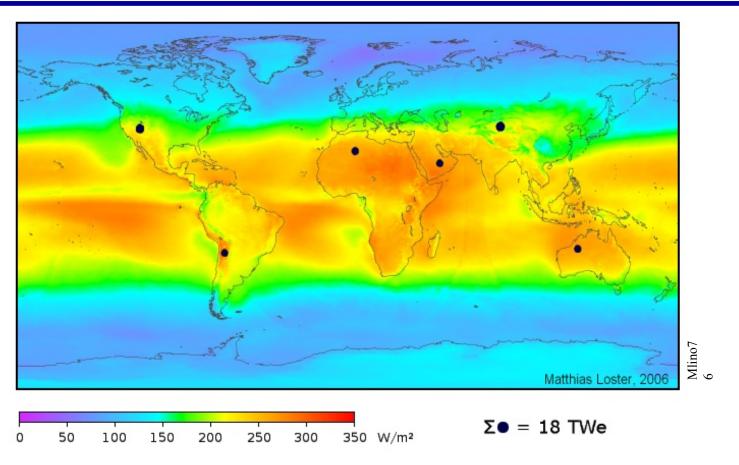
Overview



- •Who am I?
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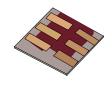
Solar radiation map

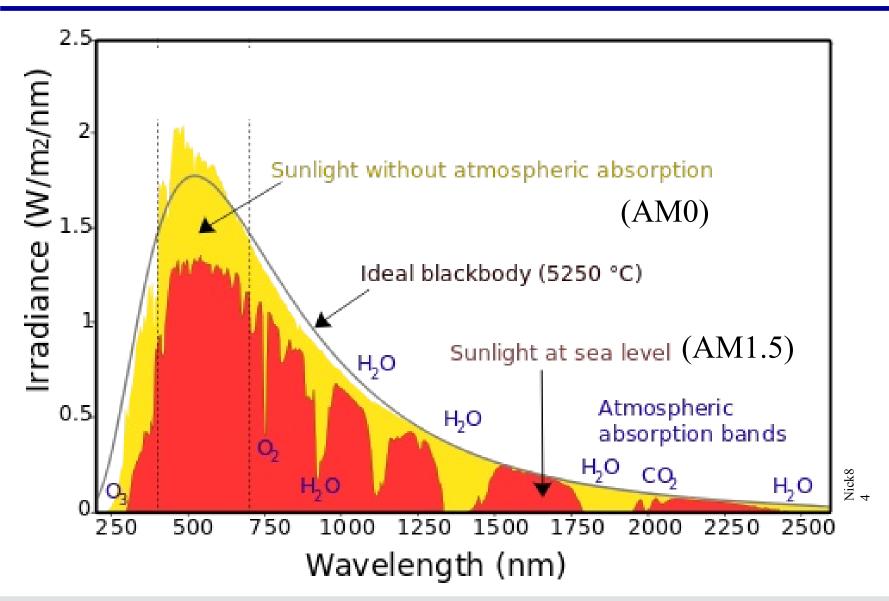




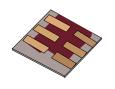
Wiki: "Solar areas defined by the dark disks could provide more than the world's total primary energy demand (assuming a conversion efficiency of 8%). That is, all energy currently consumed, including heat, electricity, fossil fuels, etc., would be produced in the form of electricity by solar cells. The colors in the map show the local solar irradiance averaged over three years from 1991 to 1993 (24 hours a day) taking into account the cloud coverage available from weather satellites."

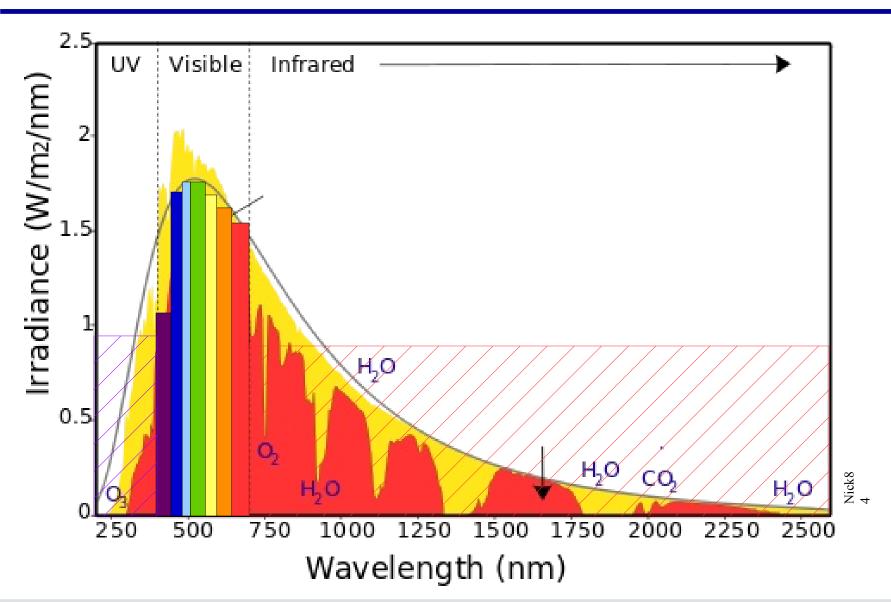
What is sunlight exactly: The solar spectrum?



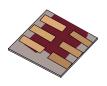


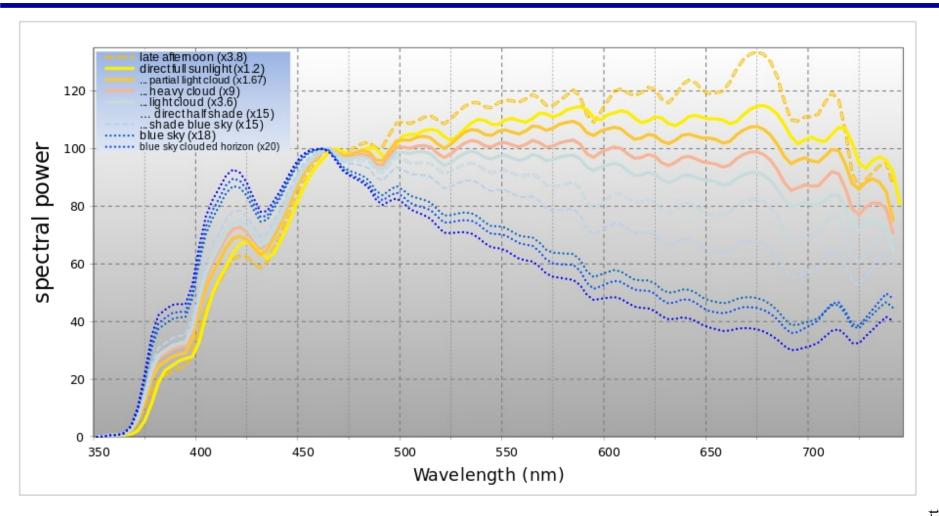
The solar spectrum





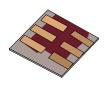
Clouds and sunlight

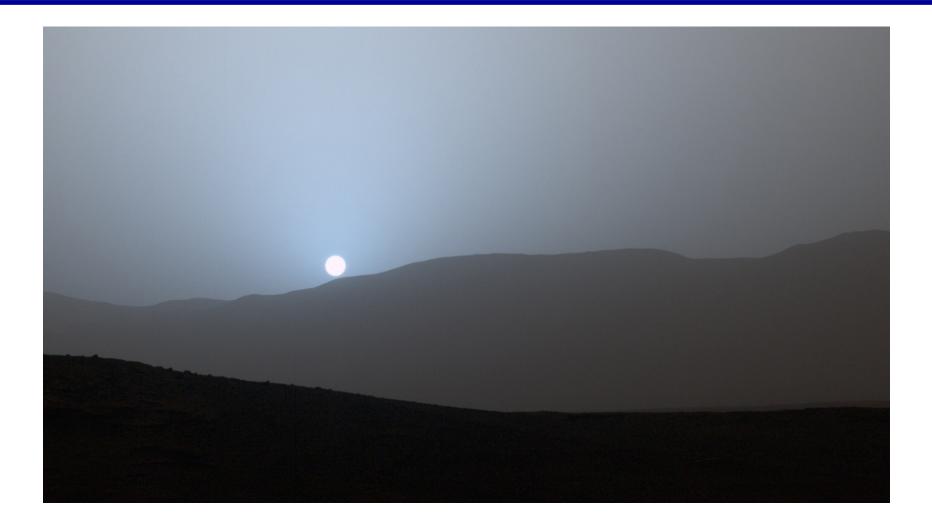




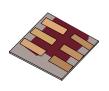
ıxbanger

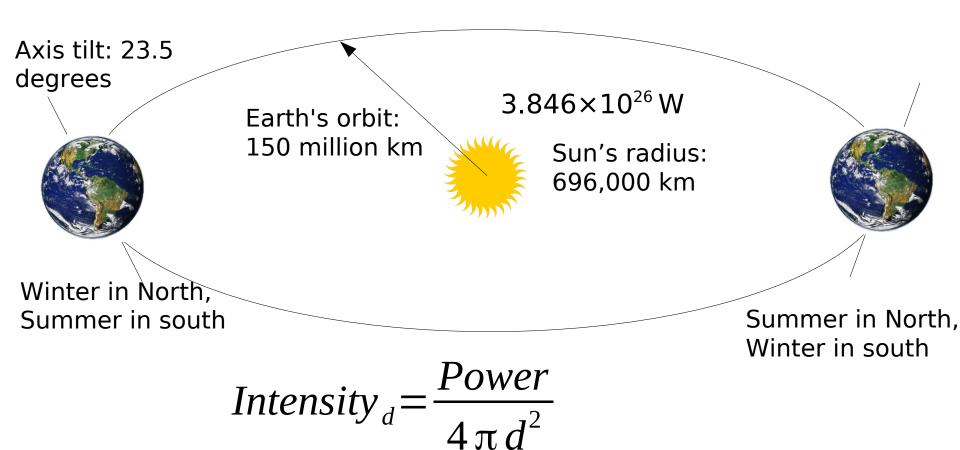
What do you think this is an image of?





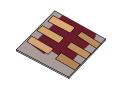
Seasonal variation in solar intensity







Class exercise





Intensity_d =
$$\frac{Power}{4\pi d^2}$$

Intensity_d =
$$\frac{3.846 \times 10^{26}}{4 \pi (150000 \times 10^{6})^{2}}$$

Intensity_d=
$$1360.2 \, \text{Wm}^{-2}$$

Overview

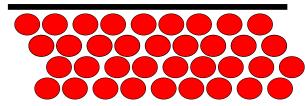


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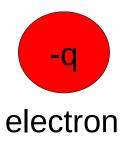
Fundamental principle of semiconductors: Let's first think about **metals**



Surface of met



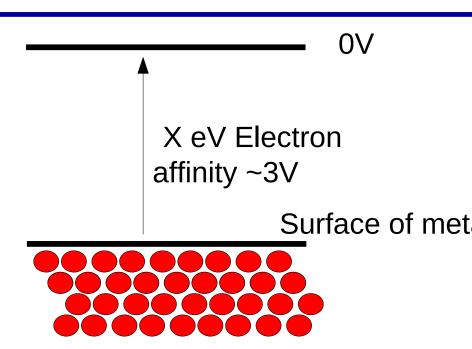
•You can think of a metal as a see of electrons.



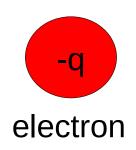
 $q=1.60217657 \times 10^{-19}$ coulombs

Fundamental principle of semiconductors: Let's first think about **metals**





- You can think of a metal as a see of electrons.
- •You need energy X (Electron affinity) to remove an electron from the surface of the metal.



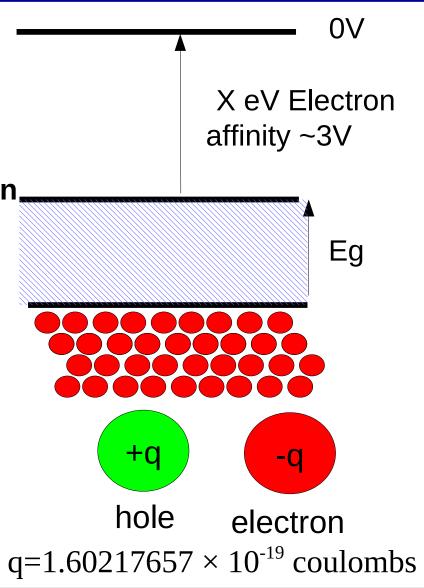
 $q=1.60217657 \times 10^{-19}$ coulombs

•Solar cells are made from a special type of material called a semiconductor.

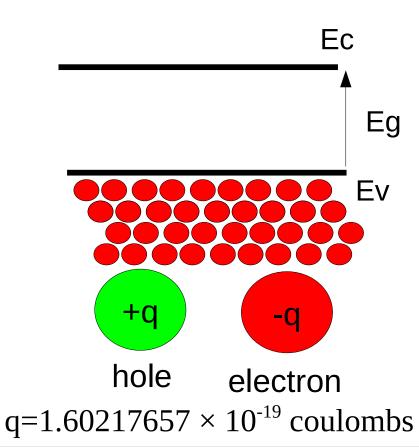


- •Solar cells are made from a special type of material called a semiconductor.
- •A semiconductors are a special because they have a **forbidden region** called the **band gap (Eg)** where no charge can exist.



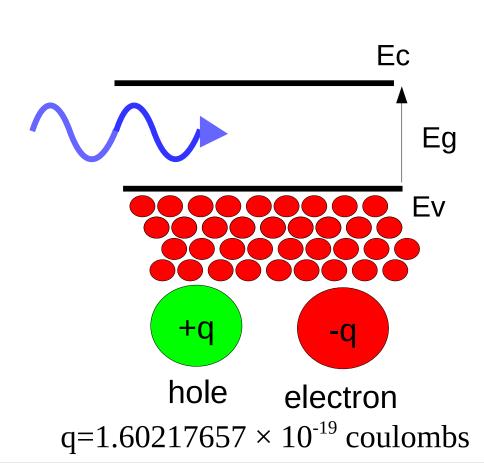


•Light can be absorbed by semiconductors.



35

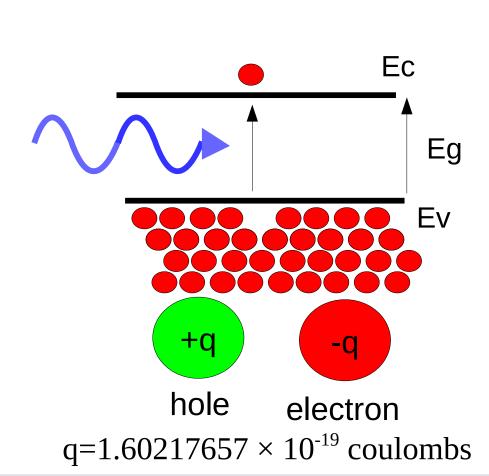
- •Light can be absorbed by semiconductors.
- •A photon enters the material.



36

Fundamental principle of semiconductors: **Semiconductors and the band gap**

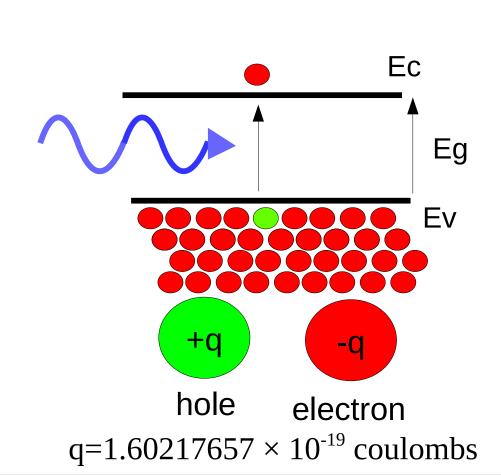
- •Light can be absorbed by semiconductors.
- •A photon enters the material.
- •And promotes an electron from **Ec** to **Ev**.



37

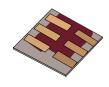
Fundamental principle of semiconductors: **Semiconductors and the band gap**

- •Light can be absorbed by semiconductors.
- A photon enters the material.
- •And promotes an electron from **Ec** to **Ev**.
- •The gap that the electron leaves is called a hole.



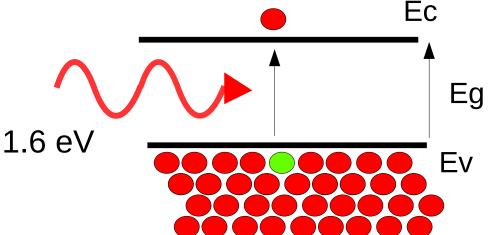
38

Absorption

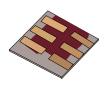


- •Silicon has a band gap of Eg 1.6 eV
- •So if a photon has exactly 1.6 eV (deep red) it will be absorbed.

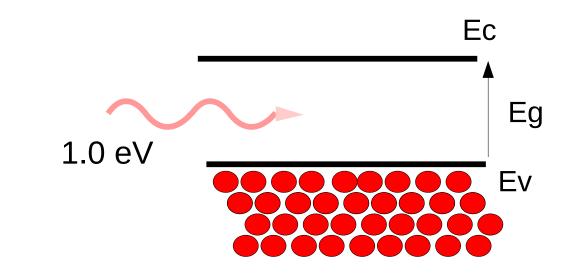




Absorption

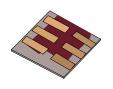


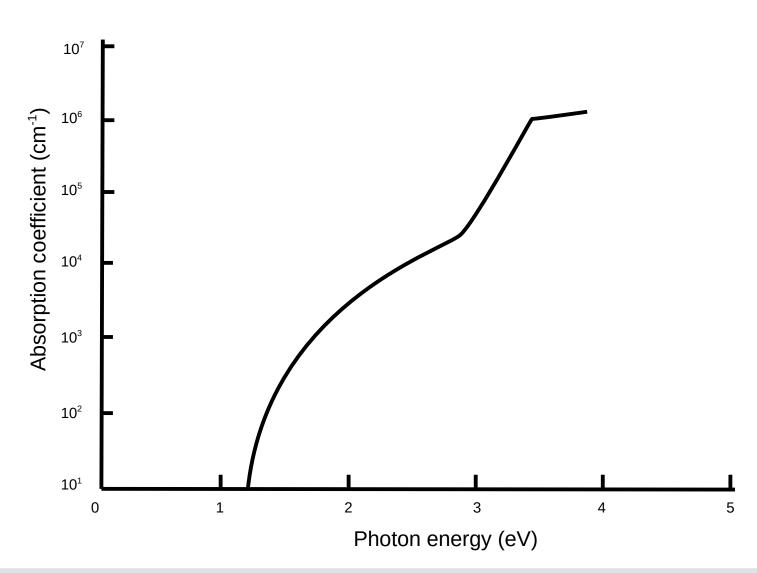
•If a photon has less than 1.6 eV of energy say (infrared red) it will not be absorbed because it does not have enough energy to promote an electron from Ev to Ec.



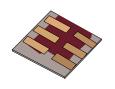


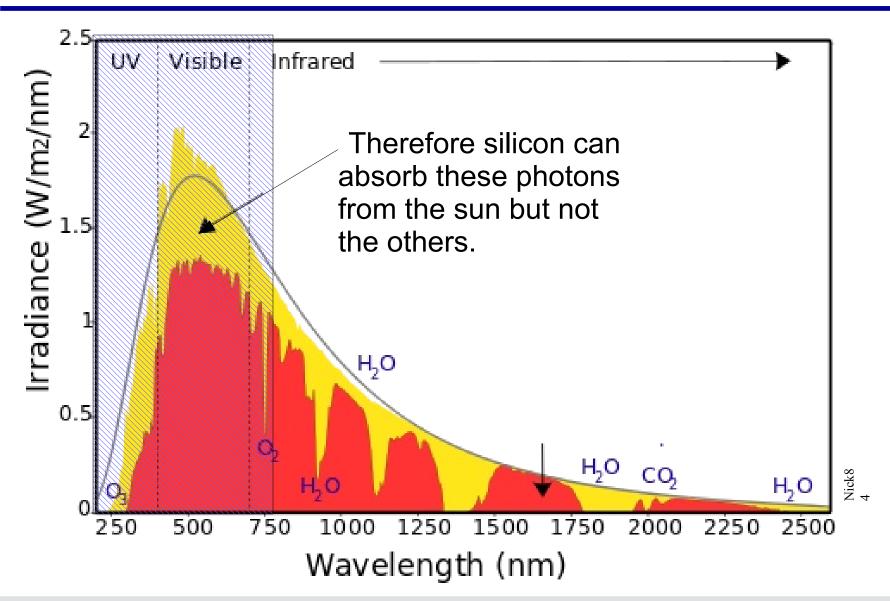
Absorption spectrum of silicon.



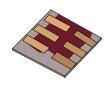


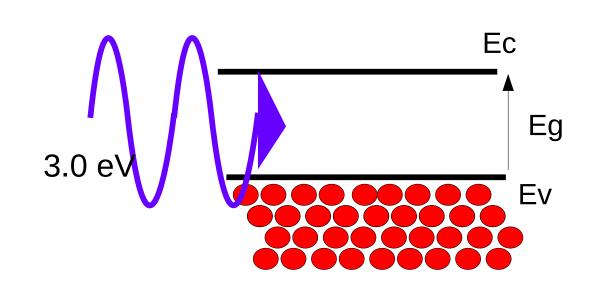
The solar spectrum



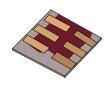


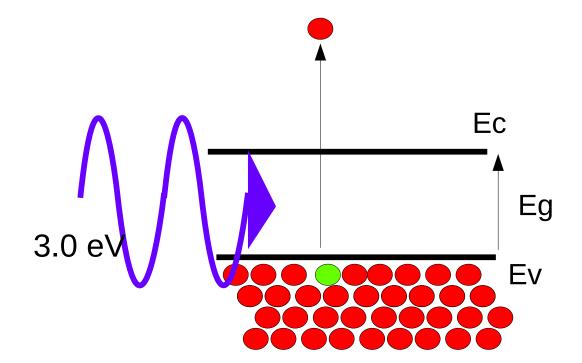
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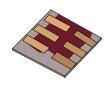


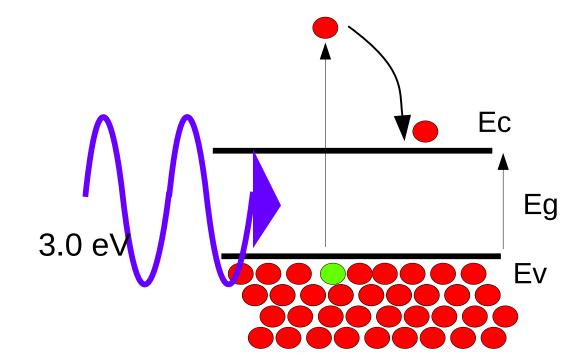
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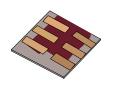


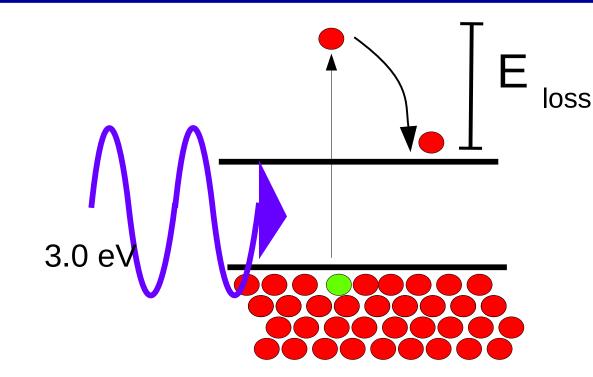
Absorption



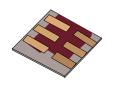


Absorption





From Energy to wavelength



Plank's constant

$$E = hf$$

$$C = f \lambda$$

$$E = h \frac{c}{\lambda}$$

$$h=6.62607004 \times 10^{-34} \text{ m}^2 \text{ kg s}^{-1}$$

 $c=3x10^8 \text{ m}^{-2}$

From Energy to wavelength



Plank's constant

$$E = hf$$

$$C = f \lambda$$

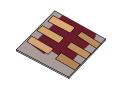
$$E = h \frac{c}{\lambda}$$

$$h=6.62607004 \times 10^{-34} \text{ m}^2 \text{ kg s}^{-1}$$

$$c=3x10^8 \text{ m}^{-2}$$

If the band gap of silicon is 1.6eV what is wavelengths of light can it absorb?

From Energy to wavelength



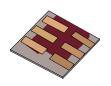
$$\lambda = h \frac{C}{E}$$
 h=6.62607004 × 10⁻³⁴ m² kg s⁻¹ c=3x10⁸ m⁻²

$$\lambda = \frac{3 \times 10^8 * 6.6 \times 10^{-34}}{(1.6 * 1.6 \times 10^{-19})}$$

$$\lambda = 7.7344 \times 10^{-07} m$$

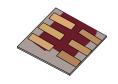
 $\lambda = 773 nm$

Overview

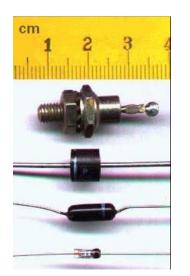


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Diodes



- •Before we talk about solar cells and light harvesting we need to know about diodes.
- •Diodes are a fundamental electrical component that form the basis for lots of classes of devices.









Power diodes

LEDs

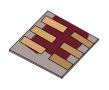
Lasers

Solar cells

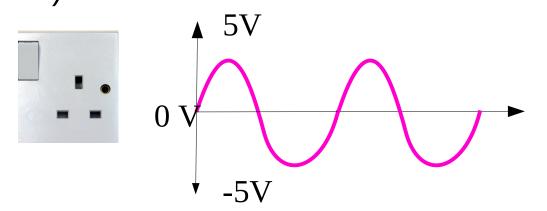
•Before we can understand solar cells we first need to understand diodes.

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Before we think about solar cells, let's think about AC voltage for a moment

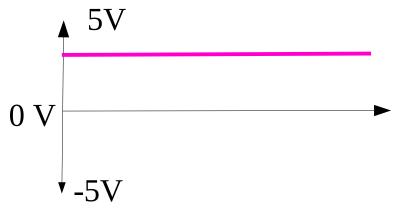


•AC switches **ON** and **OFF** again 100 times a second (50 Hz).





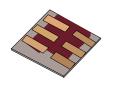
•But most chips need a steady **DC** supply to run.





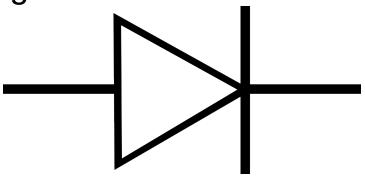
Power diodes are used to transform AC to DC.

Diode basics: Diodes for power electronics – what do they look like?

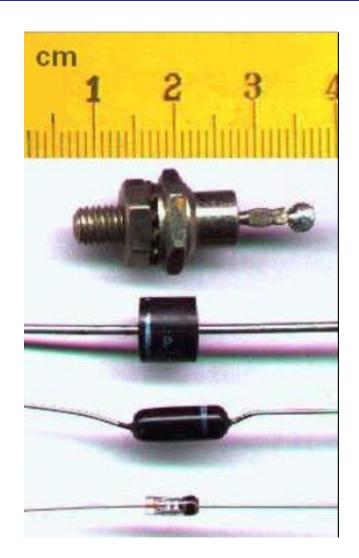


 Diodes you find in high power electronics look like this

•Little black beads with two wires sticking out.

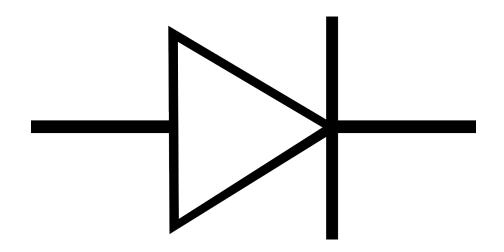


•We are looking at power didoes because they are the simplest type of diode.

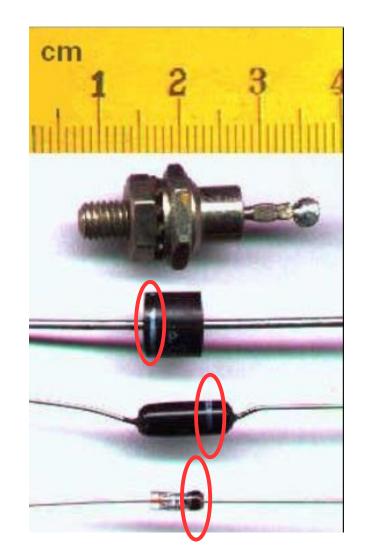


Diode basics: Notice...





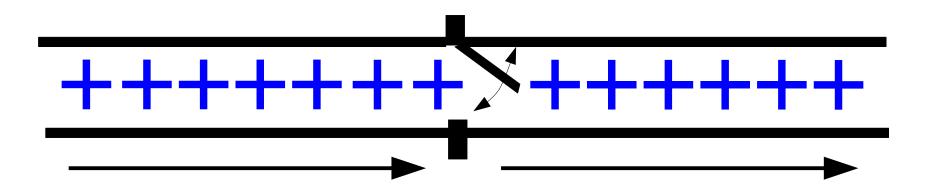
•Notice the silver bar on the end, this is the same as the bar in the diagram.

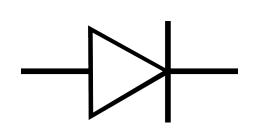


Diode basics: What do they do?



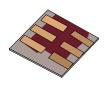
You can think of a diode as a one way valve for electrons



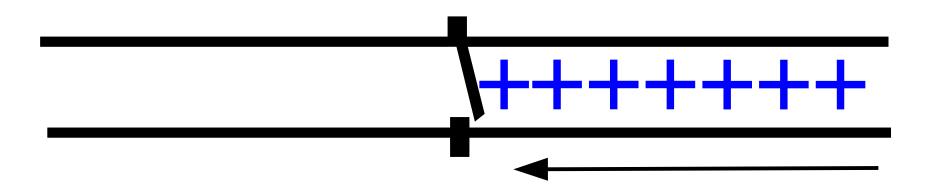


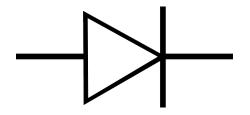
If current flows in one direction the diode will allow it to pass.

Diode basics: What do they do?



If current tries to flows in the **other direction** it will not be allowed to pass.





Diode basics: A one way cat flap....



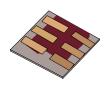
You can also think of a diode as a one way cat flap for electrons.

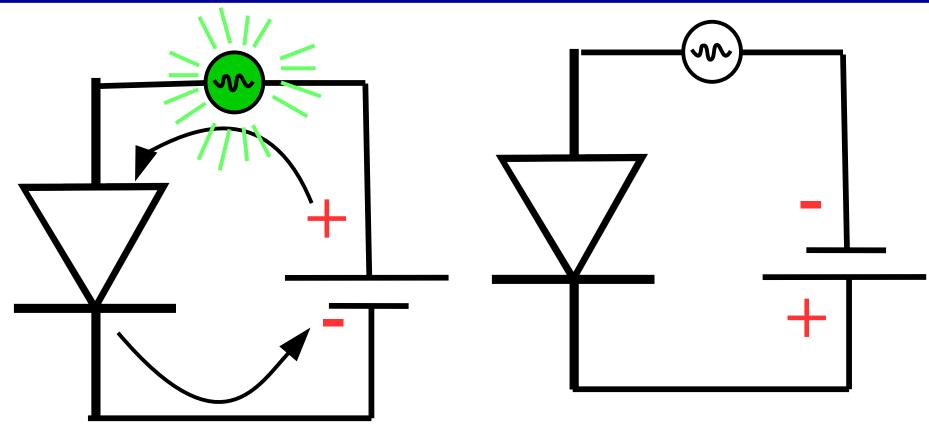
•Electrons (or the cat) are only allowed to go through one way but not the other.



Andrew Dunn

Diode basics: A diode as a one way trap door for current

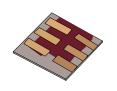


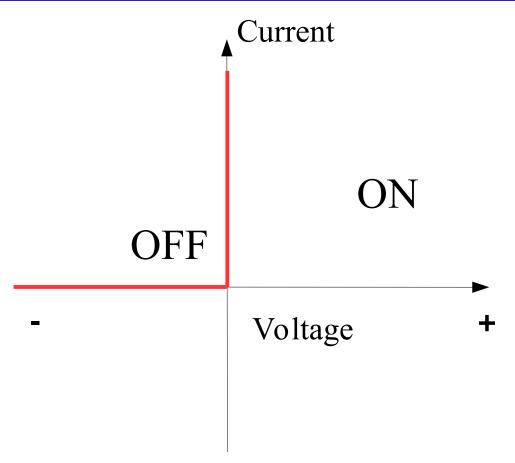


With positive applied voltage current flows

With negative applied voltage no current flows

Diode basics: What does the ideal current voltage curve look like?

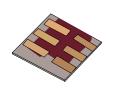


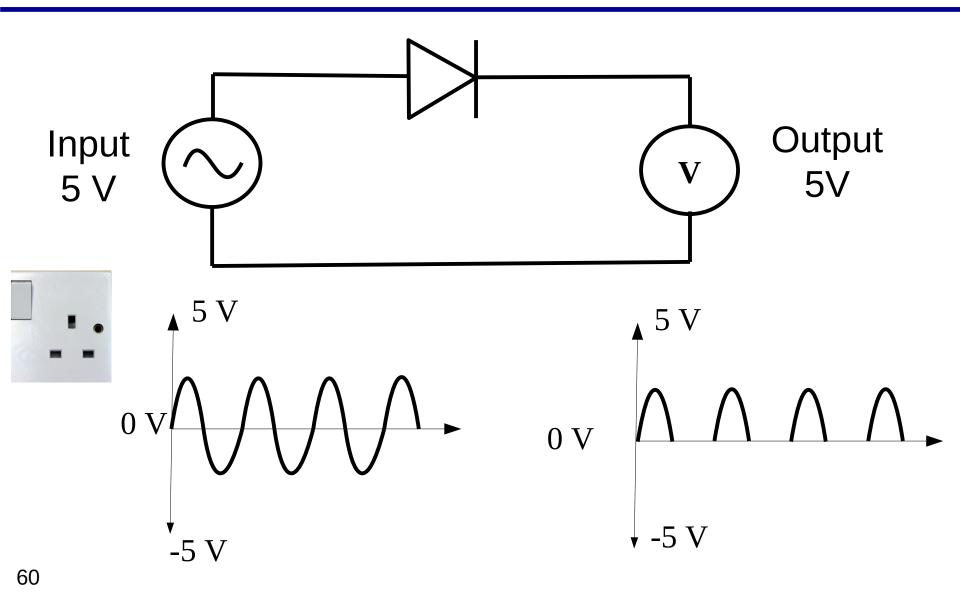


At negative voltages it blocks current

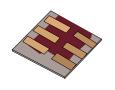
At positive voltages any amount of current can flow

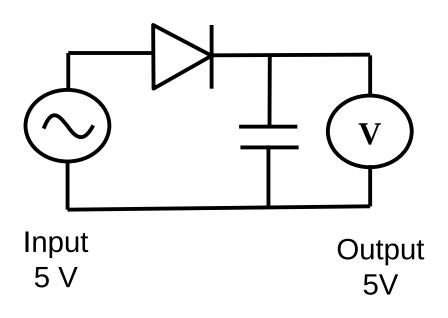
If we apply an AC signal to a diode, it will only let through the positive voltage

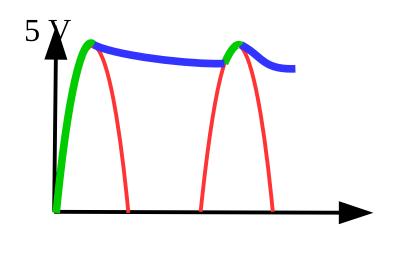




If we add a capacitor to our circuit we now get DC (sort of).



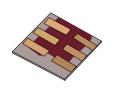


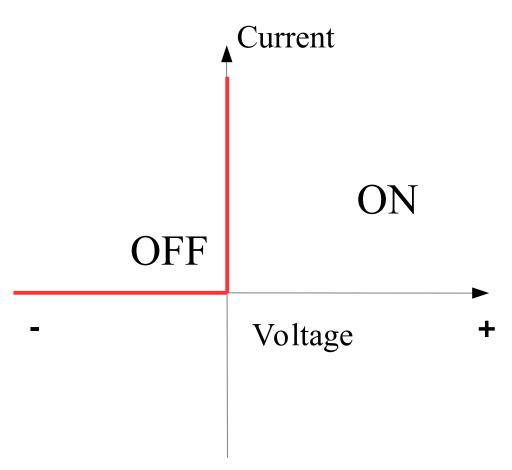


And a happy (working) MP3 player.



What we have looked at so far is the ideal diode.

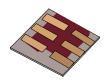


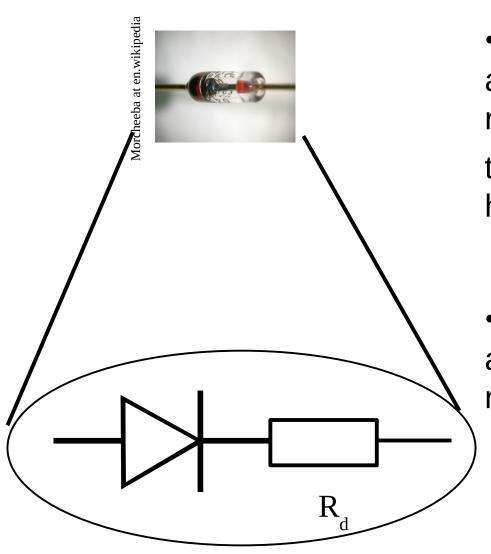


At negative voltages it blocks current

At positive voltages any amount of current can flow

A less ideal diode



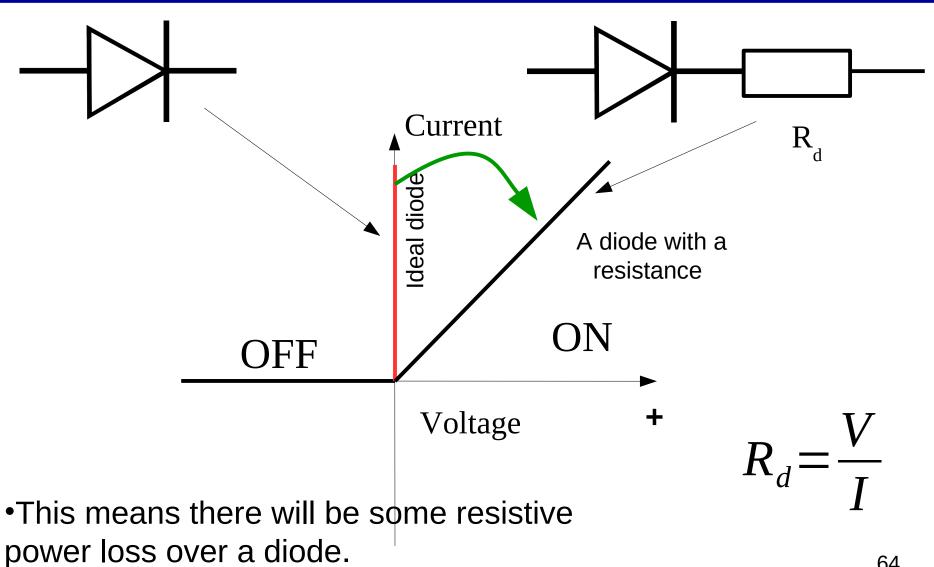


•However real diodes are like any other device they have a resistance R_d associated with them (because all things have resistance).

•So you can think of them as an ideal diode in series with a resistor.

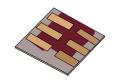
A less ideal diode



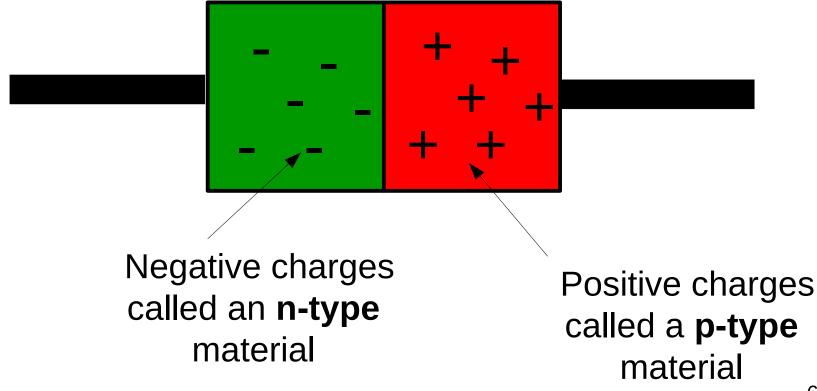


64

Furthermore.....



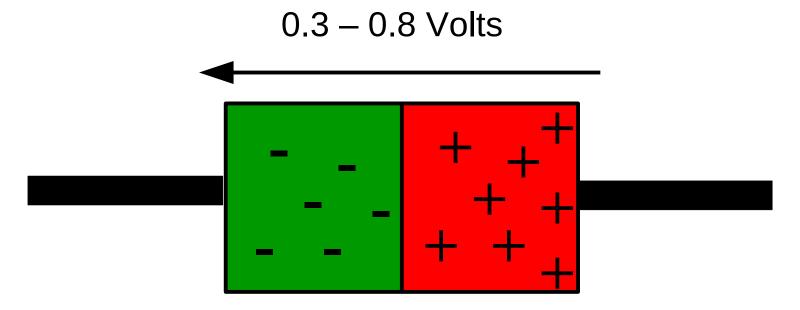
•Diodes are made of two materials, one with lots of negative (n-type) charges and one with lots of positive charges (p-type).







- •This charge in the device means that every diode produces a voltage of between 0.3 V to 0.8 V.
- This is called the built in potential.

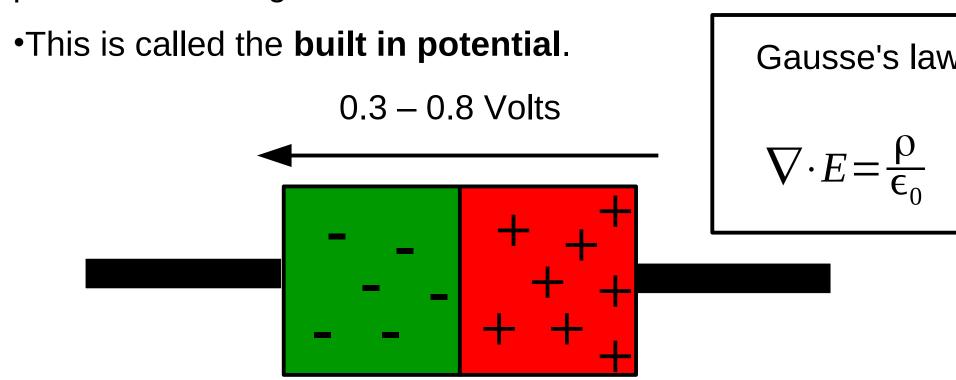


•In some applications this built in potential is a pain, in others it is really useful



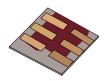


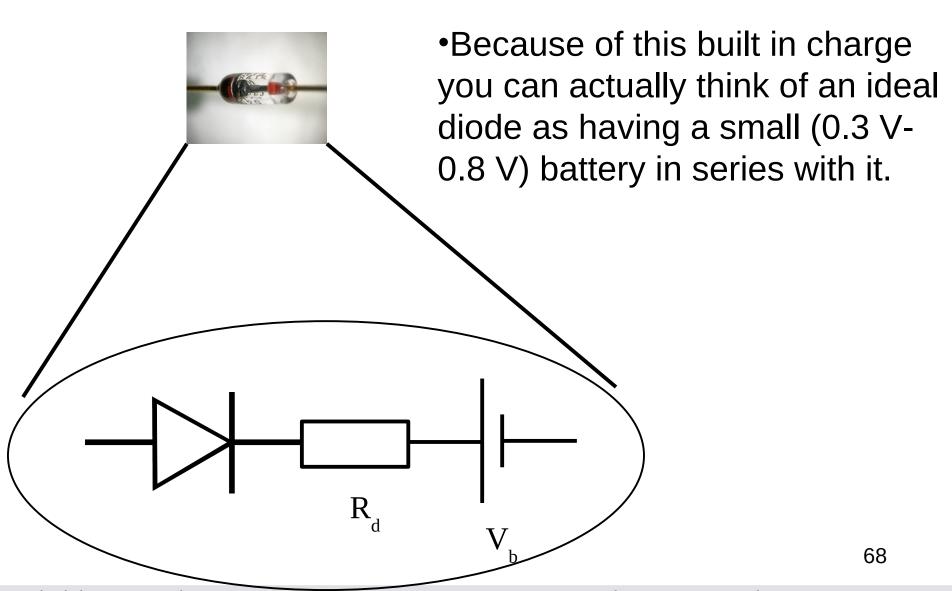
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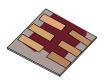
A diode with a built in potential

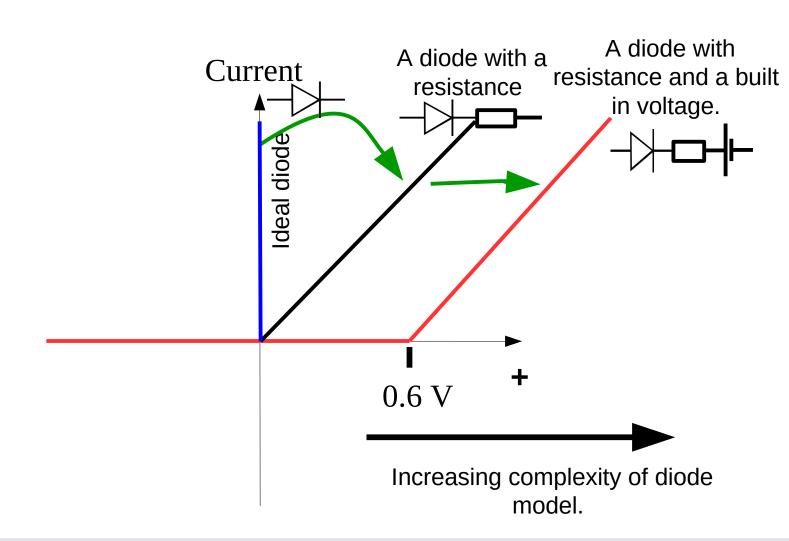




Roderick MacKenzie

The built in potential





Overview

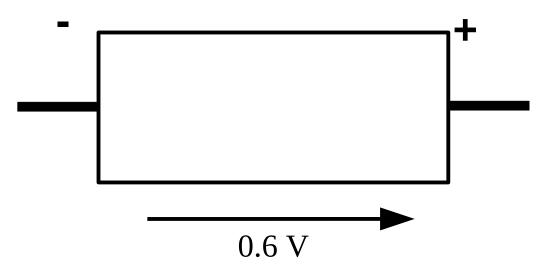


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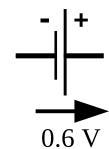
From a diode to a solar cell..



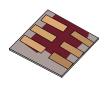
•Here is our diode again.



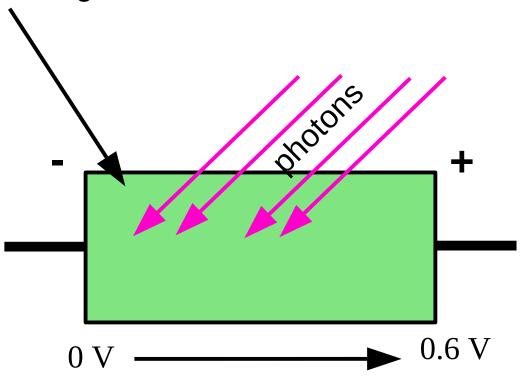
•Now let's change the material so it absorbs light.

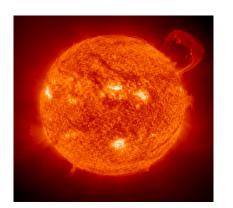


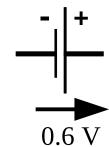
A light adsorbing diode...



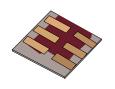
Light adsorbing material.



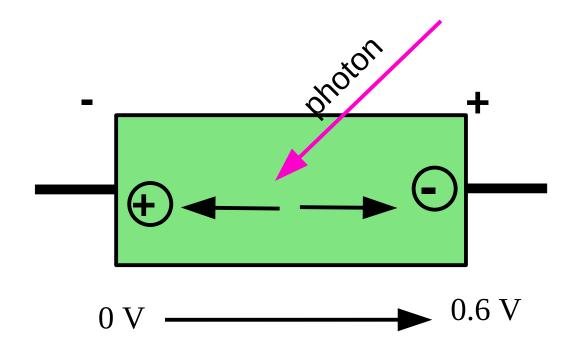




When a photon is adsorbed in a material.....



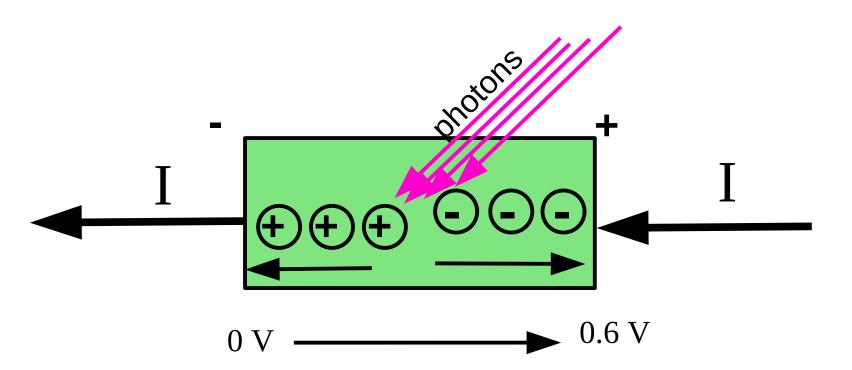
•The positive charge goes to the negative contact and the negative charge goes to the positive contact.



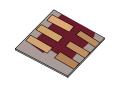
When a photon is adsorbed in a material.....

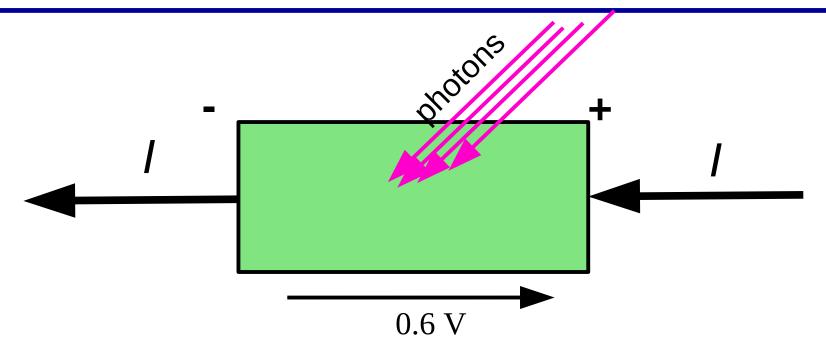


•If lots of photons hit the diode lots of positive and negative charges move to the contacts and we get current in the external circuit.



Power generation.....





Power=I*V

N=Number of photons adsorbed per second per unit area.

A=Area of solar cell.

Power=A*N*q*V

And we have a solar cell.

Exam question



a) A $0.01 \, m^2$ solar cell produces a voltage of $0.6 \, V$, it adsorbs 1x10²⁰ m⁻² photons per second if the charge on an electron is 1.6x10⁻¹⁹ coulombs how much power will it produce.

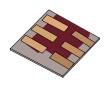
Power=A*N*q*V

Power produced by cell=???

b) How many pink **500 Watt** 'Hello Kitty' toasters would that run??



Exam question



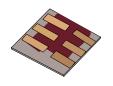
A 0.01 m^2 solar cell produces a voltage of 0.6 V, it adsorbs 1×10^{12} m^{-2} photons per second if the charge on an electron is 1.6×10^{-19} coulombs how much power will it produce.

Power=A*N*q*V

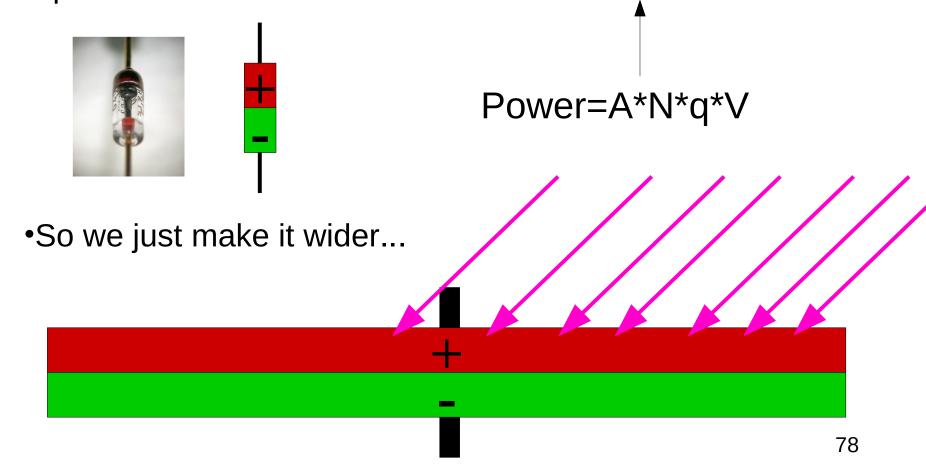
Therefore Power=0.06 W

That's not enough to run anything – let along a toaster...

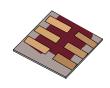
Power generation.....



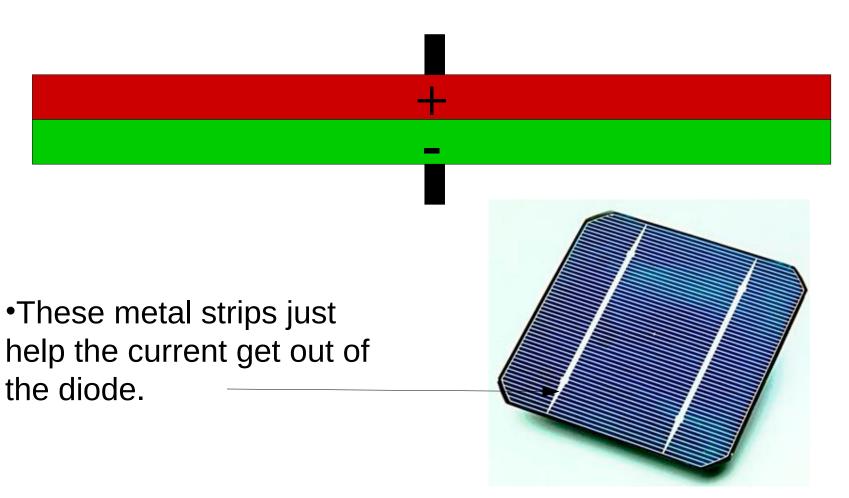
•So what we do is firstly make the diode (solar cell) as big as possible



A solar cell

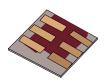


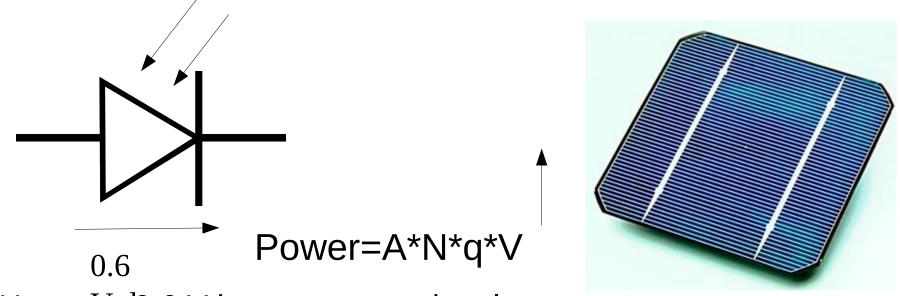
•And that is all a solar cell is – a wide diode.



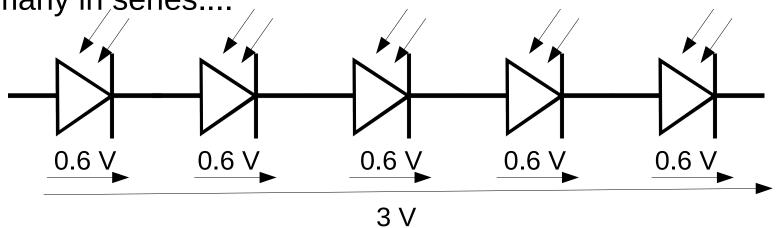
79

And then we stack lots together

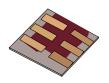




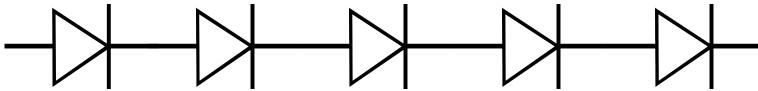
•However 66 V is not very much voltage, so we connect many in series....



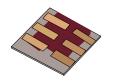
To make a solar module

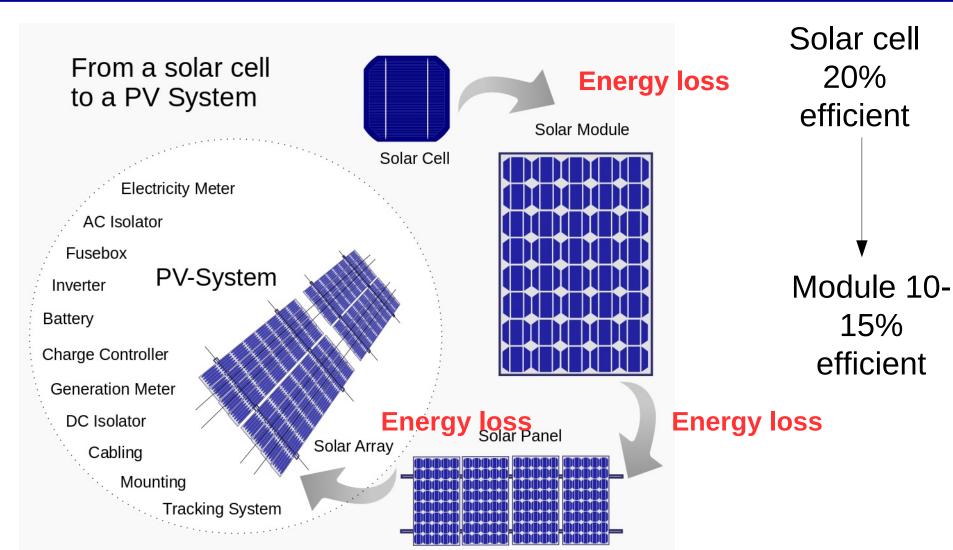






To make a solar module





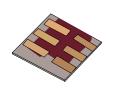
82

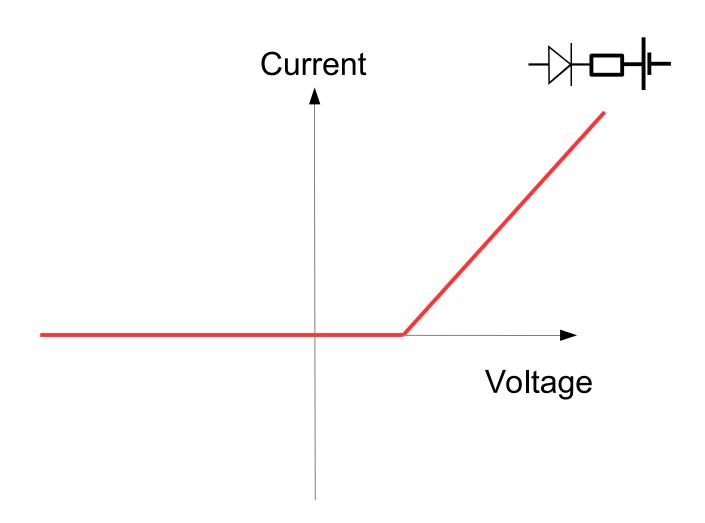
Overview

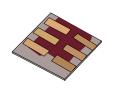


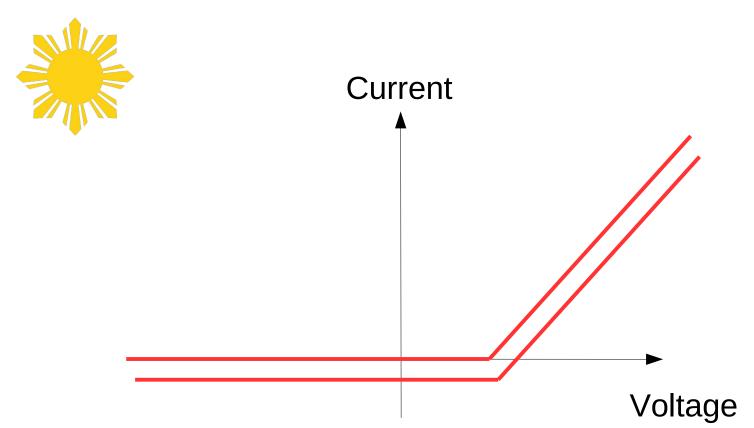
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Current Voltage curve of a diode in the light.

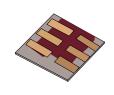


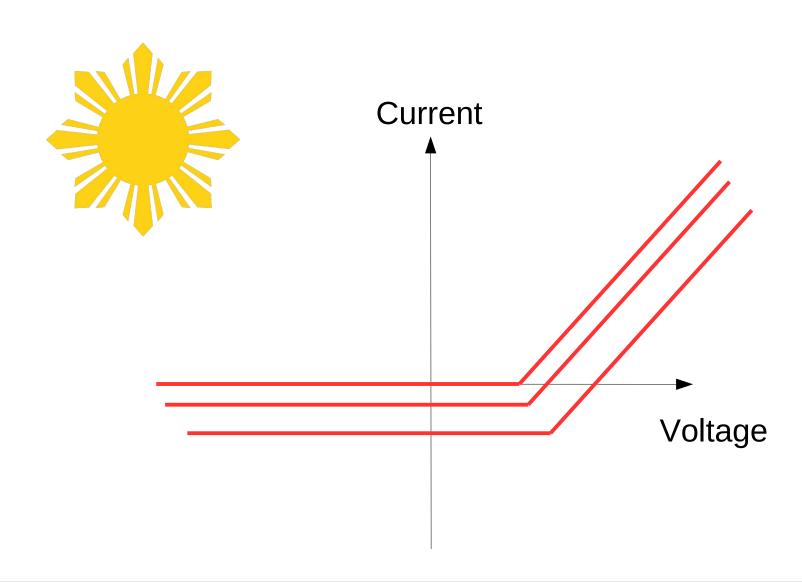


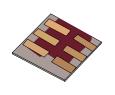


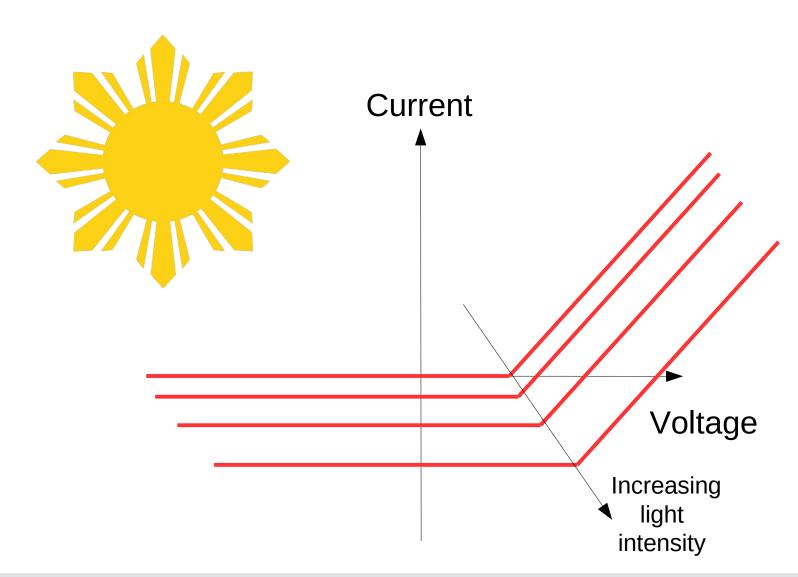


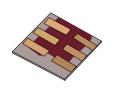
- Negative current means the current is coming out of the device rather that going into it.
- It's generating electricity.

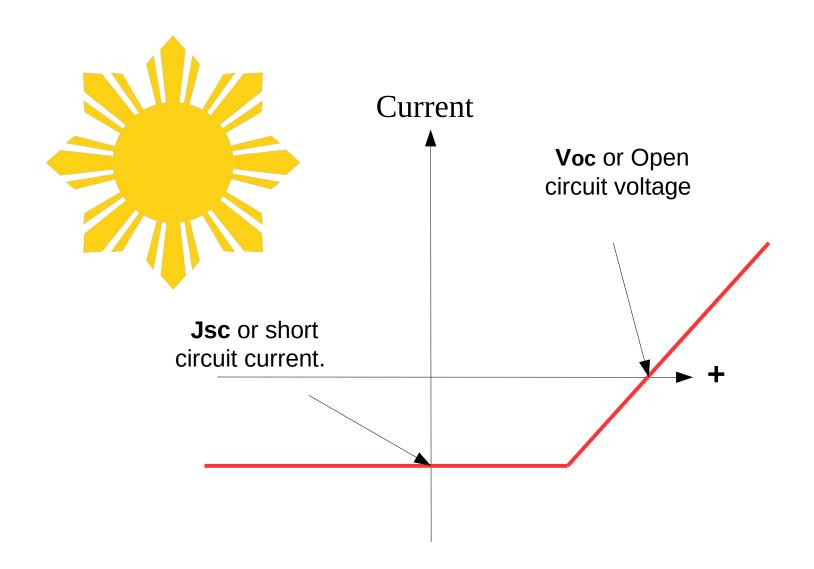




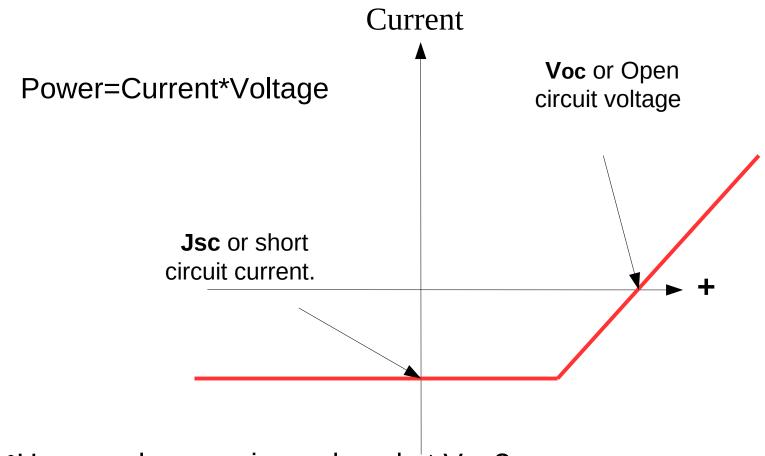






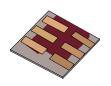






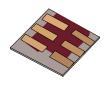
- •How much power is produced at Voc?
- •How much power is produced at Jsc?
- •Where will maximum power be produced?

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Silicon solar cells



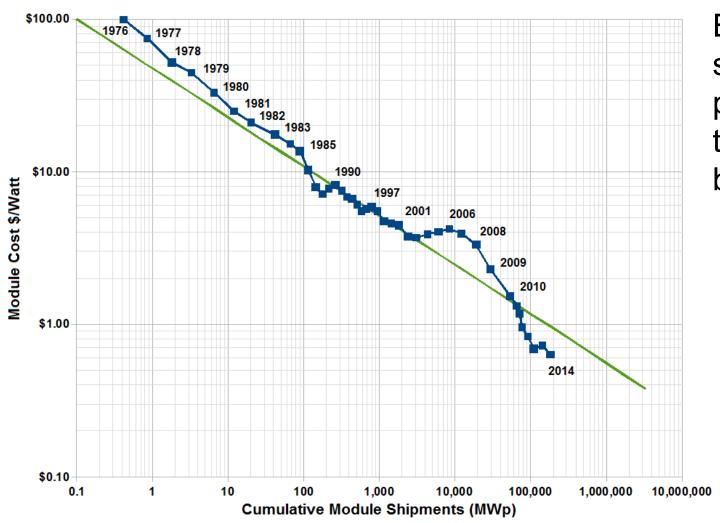


- •When you travel through the countryside and look at the roves of houses. You often see deep blue solar cells.
- •These are cells made of silicon.

- •They are about between 15-20% efficient.
- •And have a life time of between 10 and 20 years.

Swanson's law



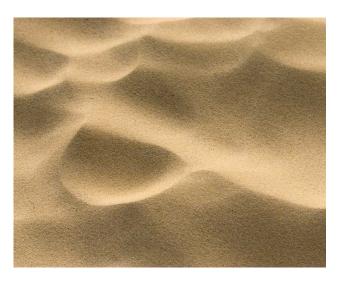


Every time solar cell production the price falls by 20%.



What's wrong with silicon?





Naturally occurring silicon



Mono-crystalline silicon



Silicon solar cell



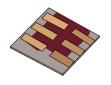
2 GJ per square meter! (553 kWh)



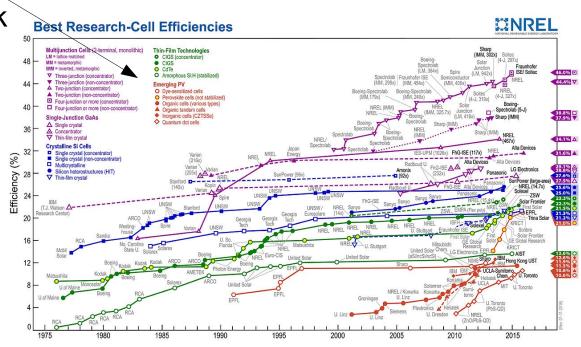
Roderick MacKenzie

https://www.oghma-nano.com, 2016

Other classes of solar cell

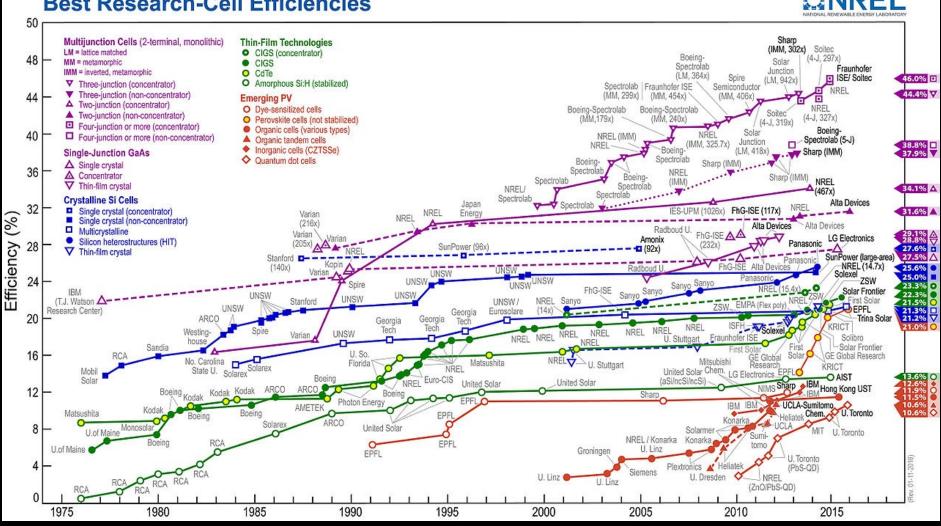


- People have been searching for alternatives to silicon solar cells for a long time
- •Below is a graph of efficiency of different types of solar cell as a function of time.
- Let's have a closer look
- •Before we look at some of these technologies in more detail.

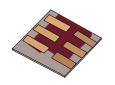


Best Research-Cell Efficiencies



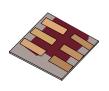


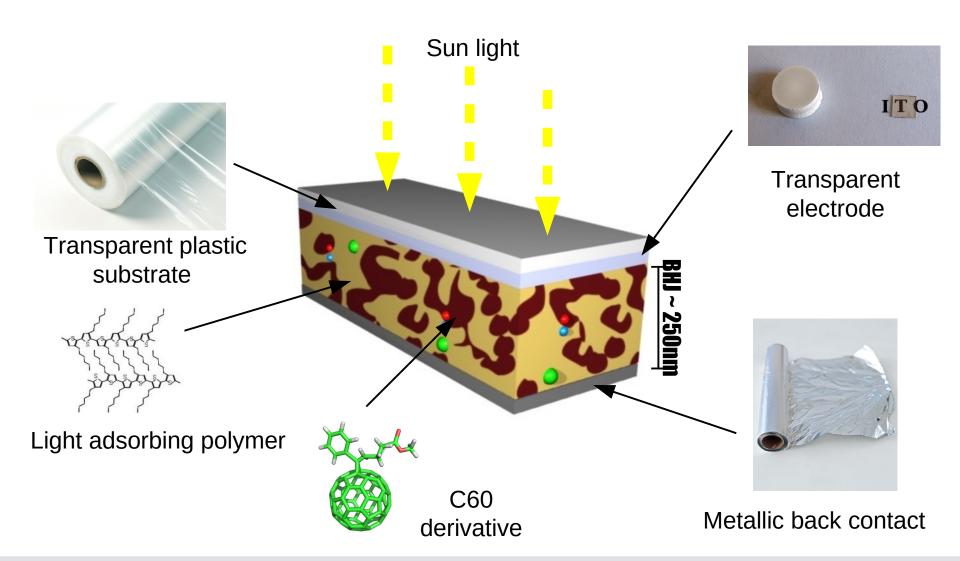
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One answer to this problem, the plastic solar cell.





What are the advantages of organic solar cells?



- •Organic molecules are cheep to make.
- •They are flexible so the cells can be easily integrated into products and buildings.

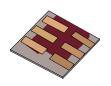






Images from www.konarka.com

Advantages of being flexible



•But most importantly:

- •Organic devices can be printed onto a plastic substrate just like newspapers are printed onto paper at (100>m/min).
- •The principle is that does not matter that they are not very efficient as they are cheep to manufacture.

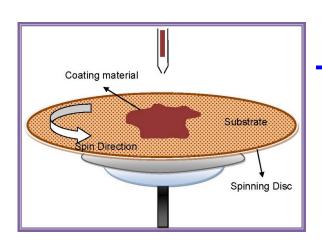




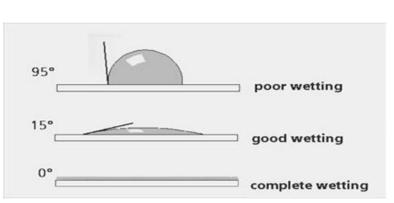
M. M. Voigt, R C.I. Mackenzie, et al. Solar Energy Materials and Solar Cells, 95, 2, 2011, pp. 731-734

From the lab to the factory: Tuning the ink parameters





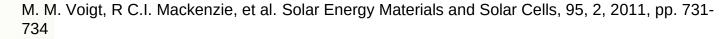
Lab scale production (1 cm² at most)



Polymer/fullerene have to be optimized for being printed too.

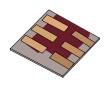


Industrial production (>100 cm²)



M. M. Voigt, R C.I. Mackenzie, et al."Gravure printing of inverted organic solar cell structures on flexible substrates" Solar Energy Materials and Solar Cells, submitted

Overview

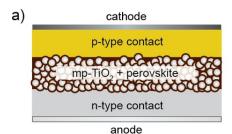


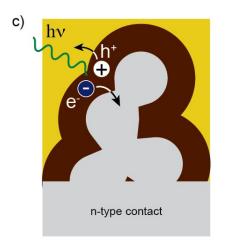
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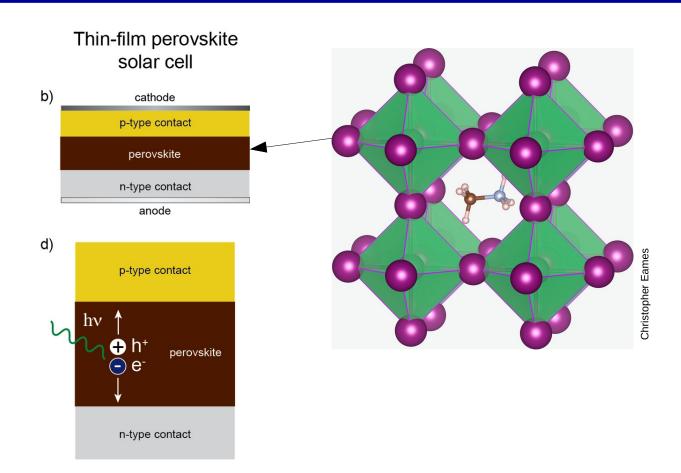
Perovskite solar cells



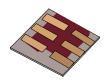
Sensitized perovskite solar cell

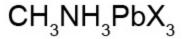


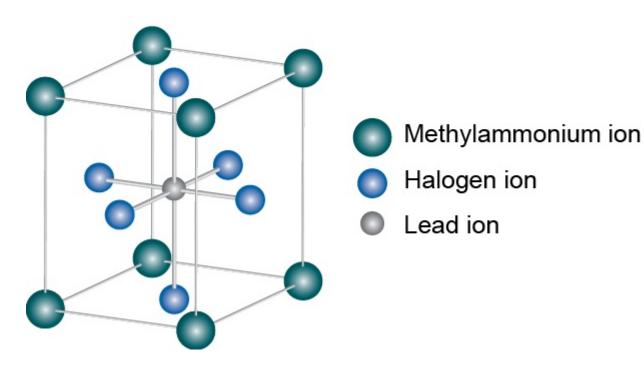




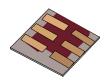
Anybody see any problems with this solar cell?



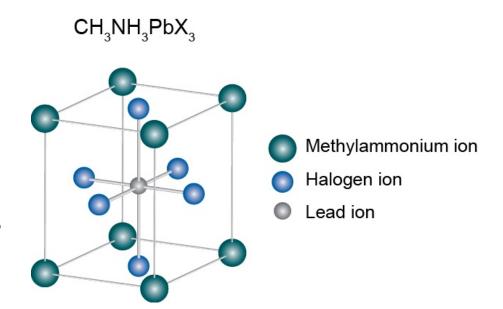




Anybody see any problems with this solar cell?

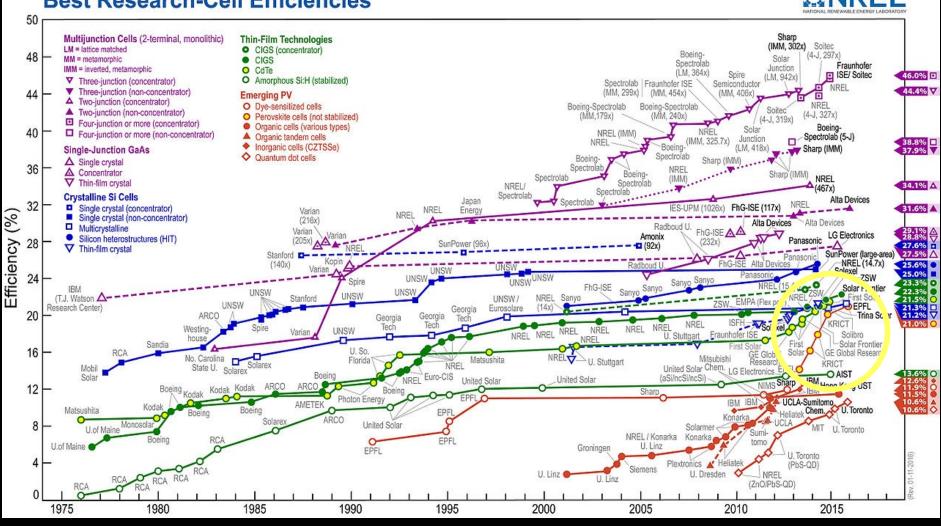


- It contains lead
- It's water soluble
- •Also, all the molecules in it can move around meaning it's a pretty unstable material.
- •Solar cells typically have a life time of hours.

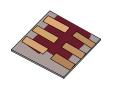


Best Research-Cell Efficiencies



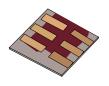


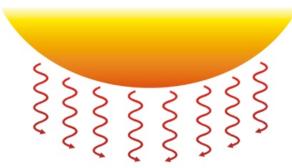
Overview

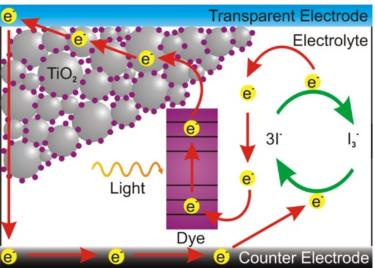


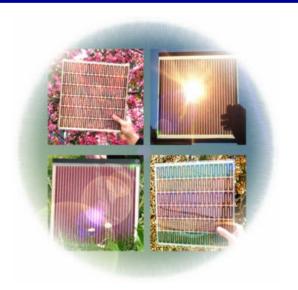
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- •Why Solar energy?
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- Absorbing sunlight in materials.
- Fundamentals of diodes
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 - Organic solar cells
 - Dye sensitized solar cells
 - Cadmium telluride solar cells
 - Multi-junction solar cells/Concentrator solar cells
- Summary

Dye sensitized solar cell











Dye sensitized solar cell



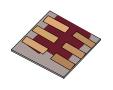
- •Efficiency never really got really high.
- •Over taken by perovskite solar cells.
- •The liquid in them was a problem.
- Never really very successful.

Overview



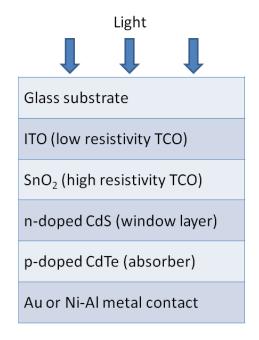
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Spotting Cadmium Telluride solar cells.

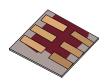




•Generally a deeper blue than silicon and don't have metallic strips on the front of them.



Problems with Cadmium Telluride solar cells.





SAFETY DATA SHEET



Environmental hazards

Hazardous to the aquatic environment, acute Category 1

hazard

Hazardous to the aquatic environment, Category 1

long-term hazard

OSHA defined hazards

Label elements

Not classified.



Signal word

Hazard statement

Danger

Toxic if swallowed. Fatal if inhaled. Suspected of causing genetic defects. May cause cancer. Suspected of damaging fertility. Suspected of damaging the unborn child. Causes damage to organs through prolonged or repeated exposure. Very toxic to aquatic life. Very toxic to aquatic life with long lasting effects.

Problems with Cadmium Telluride solar cells.





SAFETY DATA SHEET



Environmental hazards

Hazardous to the aquatic environment, acute Category 1

hazard

Hazardous to the aquatic environment, Category 1

long-term hazard

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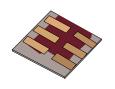
They fail the 'lick' test. Would you lick a Cadmimum Telluride solar cell?

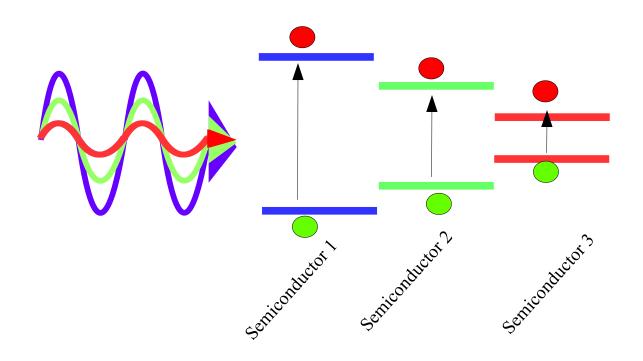
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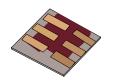
Multi-junction solar cell



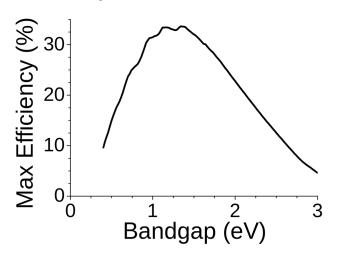


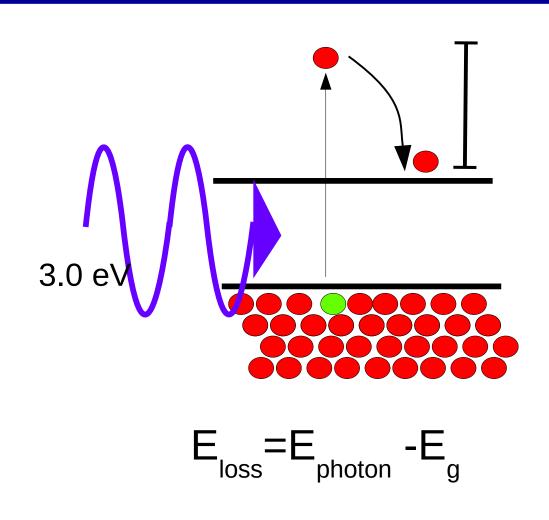
•Different parts of the solar cell are optimized absorb different parts of the sun's spectrum.

Advantages of multi-junction cells

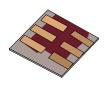


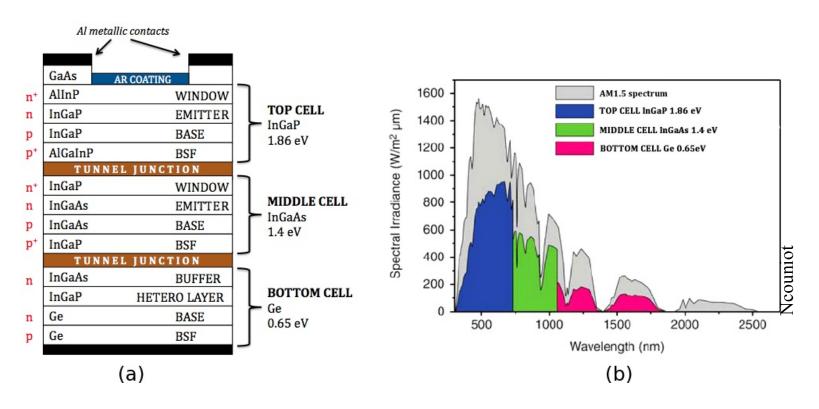
- •Multi-junction cells minimize relaxation losses.
- •Efficiency limit of a single junction cell is limited to 33.7%.
- Shockley—Queisser limit.





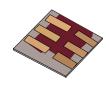
Multi-junction solar cell

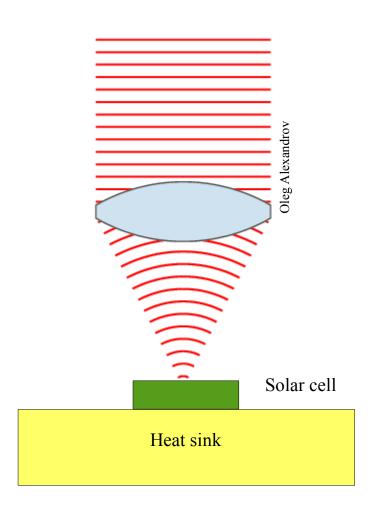




•Different parts of the solar cell are optimized absorb different parts of the sun's spectrum.

Concentrator PV

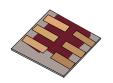






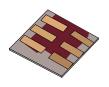
The idea is to have a highly efficient (and expensive solar cell), but use it as efficiently as possibly by putting it under 10-1000 suns. Needs cooling.

Summary

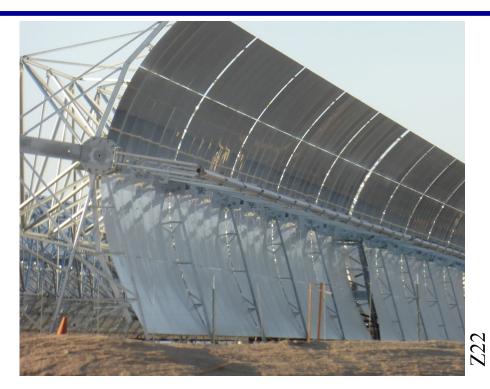


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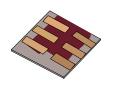
Concentrator PV

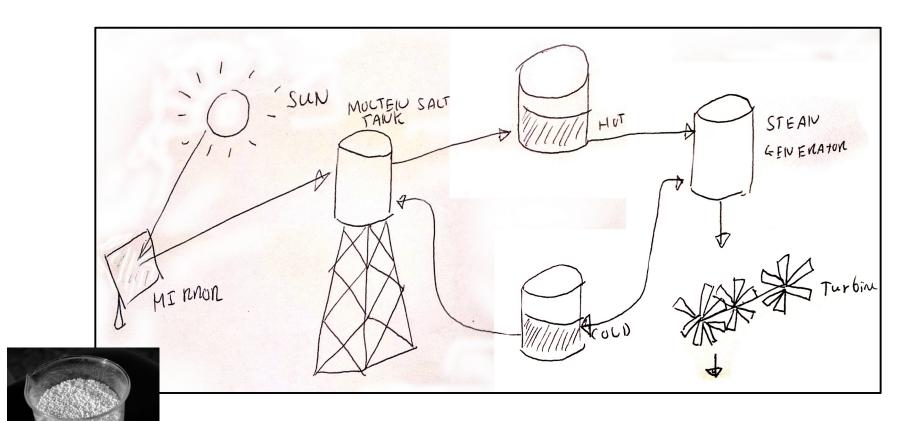




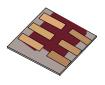


Harper Lake in California





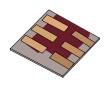
Concentrator PV





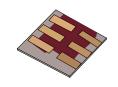
Koza1983

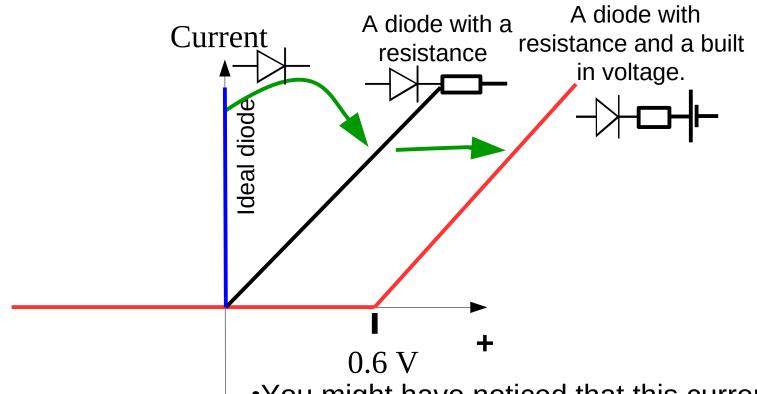
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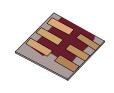
Better solar cell models

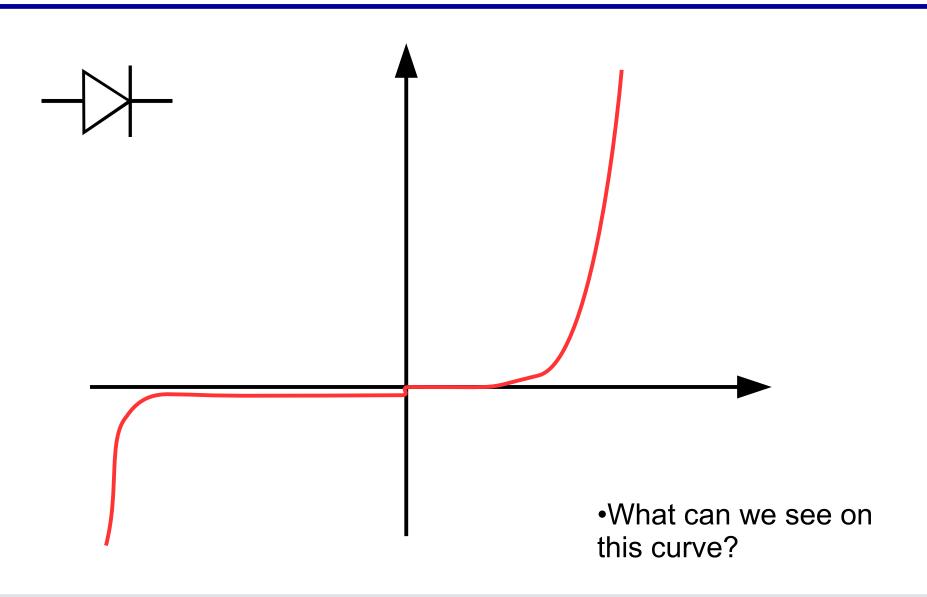




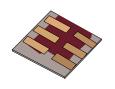
- •You might have noticed that this current voltage curve does not look like any component you have ever seen.
- •Let's look at a more realistic diode curve.

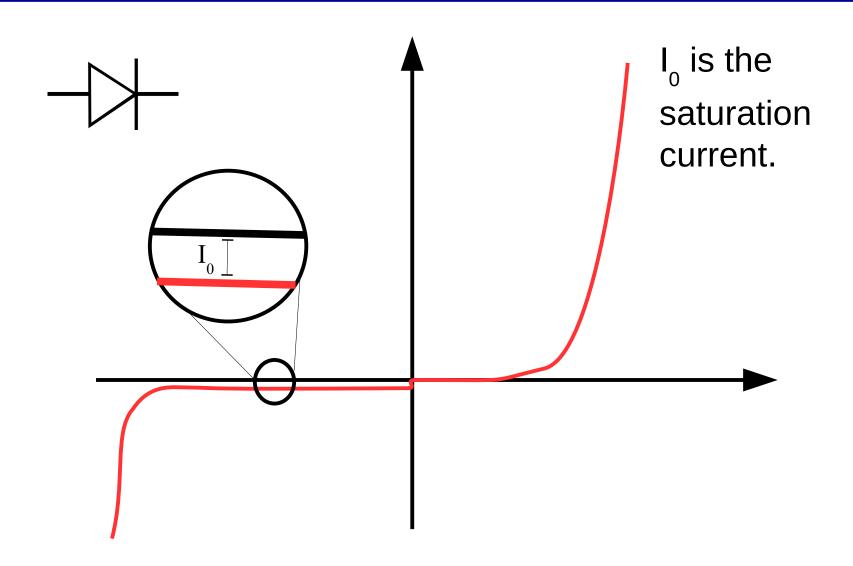
A more real diode curve



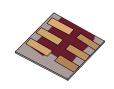


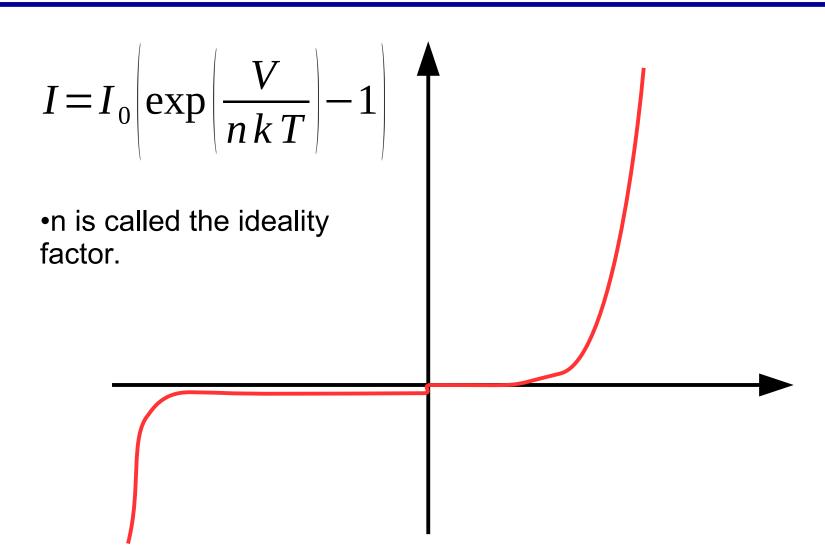
A real diode curve



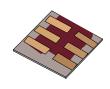


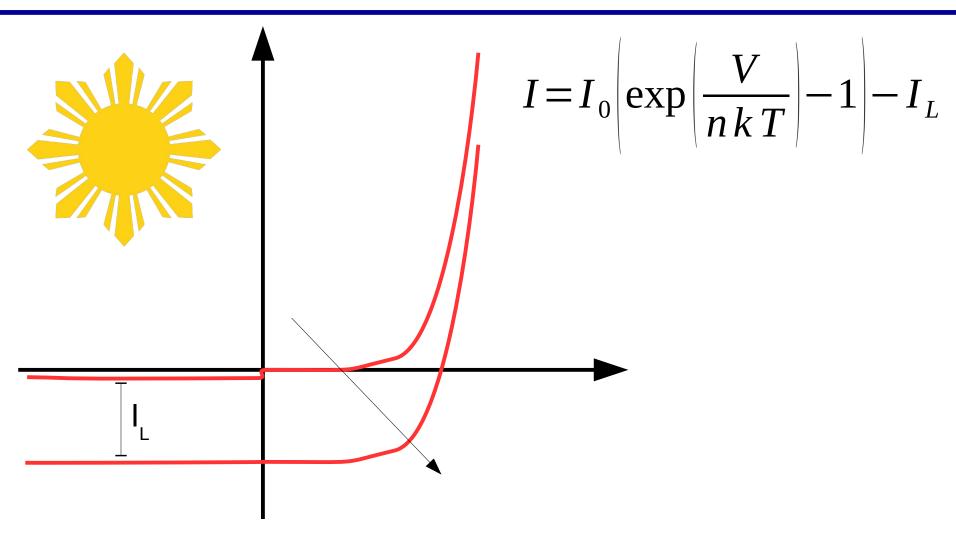
A better ideal diode model.



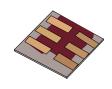


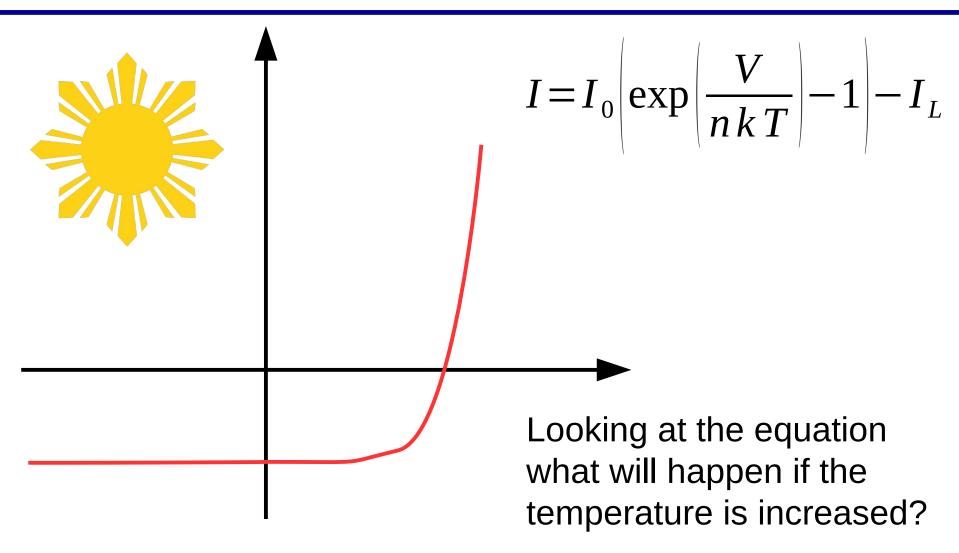
The diode equation in the light.



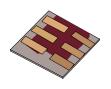


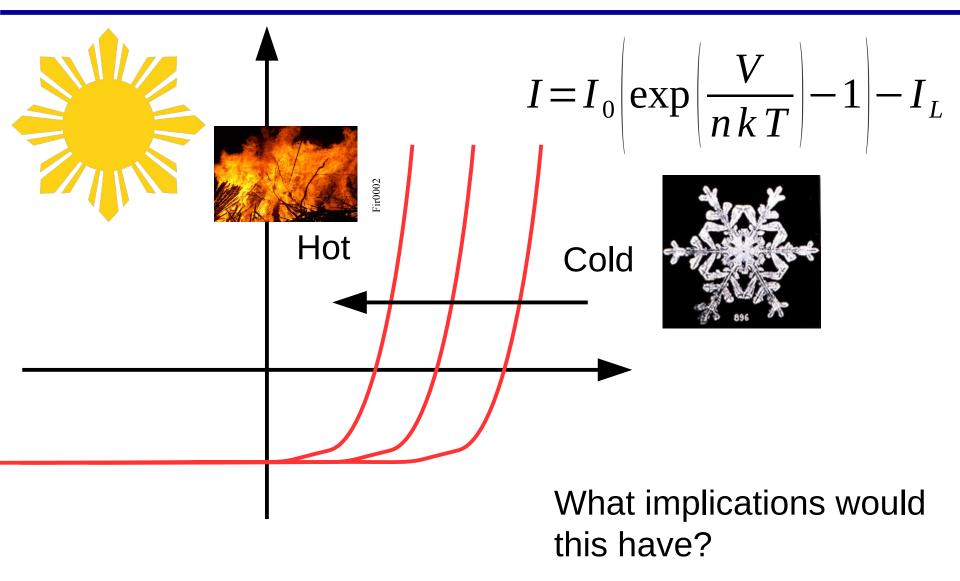
The diode equation in the light.





The diode equation in the light.

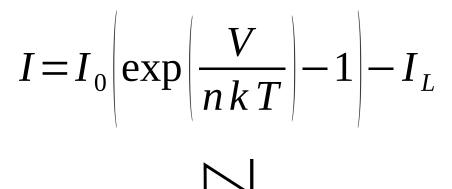


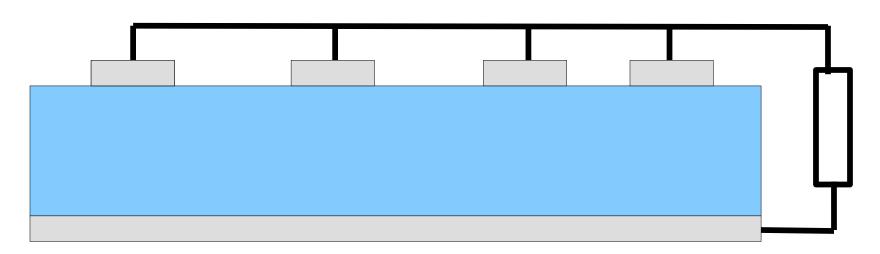


The ideal diode equation

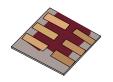


- •This equation is for an ideal diode with no resistance. However in a real solar cell there will be:
- Series resistance
- And shunt resistance





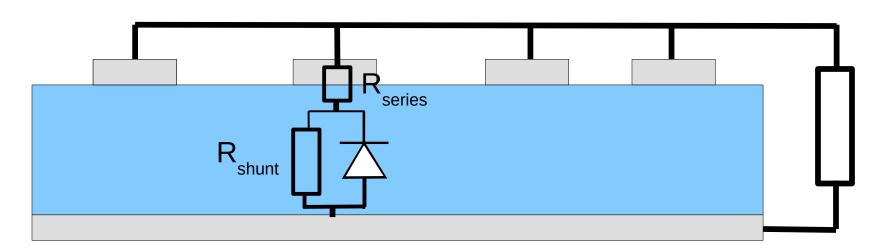
The ideal diode equation



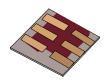
•This equation is for an ideal diode with no resistance. However in a real solar cell there will be:

$$I = I_0 \left| \exp \left| \frac{V}{n k T} \right| - 1 \right| - I_L$$

- Series resistance
- And shunt resistance



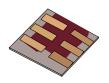
A better equivalent circuit



- Series resistance (1-10 Ohm)
- And shunt resistance (1 M Ohm)

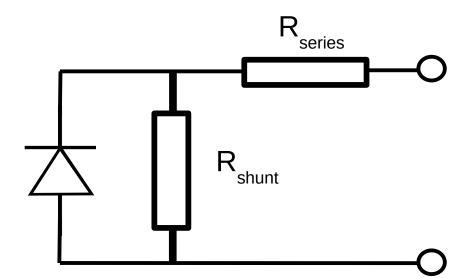
$$I = I_0 \left| \exp \left| \frac{V}{nkT} \right| - 1 \right| - I_L$$

A better equivalent circuit



Derive non-ideal diode equation

$$I = I_0 \left| \exp \left| \frac{V}{nkT} \right| - 1 \right| - I_L$$

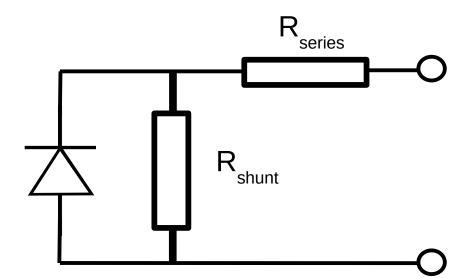


Dark JV curve

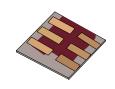


Derive non-ideal diode equation

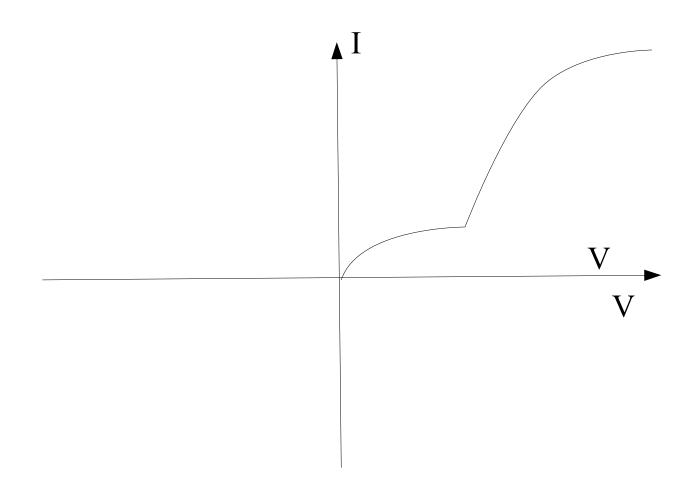
$$I = I_0 \left| \exp \left| \frac{V}{nkT} \right| - 1 \right| - I_L$$



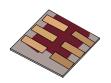
Dark JV curve



Derive non-ideal diode equation



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